# Regioselective Ortho Olefination of Aryl Sulfonamide via Rhodium-Catalyzed Direct C−H Bond Activation

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**S** Supporting Information



ABSTRACT: Rh(III)-catalyzed ortho C−H olefination of aryl sulfonamide directed by the SO<sub>2</sub>NHAc group is reported. This oxidative coupling process is achieved highly efficiently and selectively with a broad substrate scope. The reactions of Ntosylacetamide with acrylate esters afford ortho-alkenylated benzofused five-membered cyclic sulfonamides, whereas styrenes provide the direct diolefination products.

## ■ INTRODUCTION

Transition-metal-catalyzed C−H functionalization of arenes represents a step-economical and waste-reducing process, because no prior functionalization of the substrate is necessary.<sup>1</sup> Its synthetic applications have been well identified in the fields of organic synthesis, medicinal chemistry, and material[s](#page-8-0) science.<sup>2</sup> In this regard, the Fujiwara–Moritani reaction<sup>3</sup> oxidative olefination of normally inert aryl C−H bonds-is an attracti[v](#page-8-0)e alternative to the traditional Mizoroki−[He](#page-8-0)ck reaction.<sup>4</sup> Among such transformations, palladium(II) complexes have been extensively explored and widely used due to their re[m](#page-8-0)arkable activities.<sup>5</sup> Later, after the pioneering work of Satoh and Miura, rhodium $(III)^6$  and ruthenium $(II)^{7,8}$  species were also found to exhibit [h](#page-8-0)igh catalytic activity in the oxidative alkenylation reactions. In the[se](#page-9-0) reactions, the u[tilit](#page-9-0)y of a neighboring directing group has proven to be one of the most promising strategies for ensuring regioselective C−H activation.<sup>9</sup>

Sulfonamide functional groups have long been acclaimed as esse[nt](#page-9-0)ial structural motifs presented in drugs such as darunavir, celecoxib, and gliclazide.<sup>10</sup> Meanwhile, sultams (cyclic sulfonamides) have emerged as privileged structures in drug discovery due to their d[ive](#page-9-0)rse biological properties.<sup>11</sup> In particular, a number of benzofused sultams have shown promising bioactivity for treating disorders of the brain<sup>[12](#page-9-0)</sup> and exhibit broad inhibitory properties against a variety of enzymes (Figure 1).<sup>13</sup> Recent studies reveal that the sulfonamide [gr](#page-9-0)oup can act as a directing group in promoting C−H activation; however, [on](#page-9-0)ly very limited examples have been documented.<sup>14,15</sup> In 1998, Miura demonstrated the olefination reaction of N-(2′-phenylphenyl)benzenesulfonamides with acrylate [este](#page-9-0)rs via cleavage of the C−H bond at the 2′ position.15a In 2011, Li reported a rhodium(III)-catalyzed C−H



Figure 1. Biologically active molecules containing the sulfonamide or sultam substructure motif.

olefination of N-(1-naphthyl)sulfonamides at the peri position.<sup>15b</sup> Later, this group described a Rh(III)-catalyzed oxidative coupling between N-allyl sulfonamides with activated olefi[ns.](#page-9-0)15c For these substrates, the C−H bonds of arenes connected to sulfonamide sulfur atoms are totally inert under the re[acti](#page-9-0)on conditions (Scheme 1). With the assistance of  $SO_2NHC_6F_5$  as a directing group, Yu reported Pd-catalyzed sulfonamide C−H olefination. T[he](#page-1-0) reactions only produce direct alkenylation products, irrespective of the alkenes employed (Scheme 1a).<sup>15d</sup> As a continuation of our interest in Ru- and Rh-catalyzed C−H functionalization,<sup>16</sup> we herein disclose our recent [de](#page-1-0)ve[lop](#page-9-0)ment of Rh-catalyzed regioselective oxidative C−H olefination using SO<sub>2</sub>NHAc a[s a](#page-9-0) directing

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## <span id="page-1-0"></span>Scheme 1. Sulfonamide Group Directed Oxidative Olefination



group. The reactions afford ortho-alkenylated benzofused sultam analogues and diolefinated aryl sulfonamides with good yields and a broad reaction scope (Scheme 1b).

## ■ RESULTS AND DISCUSSION

We initiated our study with the oxidative olefination of readily available N-tosylacetamide  $(1a)$  and ethyl acrylate  $(2a)$ . After a considerable number of experiments (Table 1),<sup>17</sup> we found that product 3a, which was formed by dialkenylation at both ortho positions followed by intramolecular hydr[oam](#page-9-0)ination, was obtained in 93% yield in the presence of  $[Cp*RhCl<sub>2</sub>]$ <sub>2</sub> (3.0 mol %) and  $Cu(OAc)<sub>2</sub>·H<sub>2</sub>O$  (2.0 equiv) in toluene at 100 °C under an Ar atmosphere for 24 h (entry 1, Table 1). No monoalkenylated cycloadduct was observed. The structure of 3a was confirmed by  ${}^{1}H$  and  ${}^{13}C$  NMR analysis and mass spectrometry. The solvents DME and  $CH<sub>3</sub>CN$  exhibited negative results (entries 2 and 3). A change of oxidant to AgOAc,  $Ag_2CO_3$ , or  $O_2$  led to a low yield or a trace amount of the desired product (entries 4−6). Reducing the amount of  $[Cp*RhCl<sub>2</sub>]$ <sub>2</sub> from 3.0 to 2.0 mol % provided 3a in 82% yield (entry 7). It is interesting to find that when switching the ratio of 1a and 2a to 1:1 and 2:1, the reactions still gave dominantly the dialkenylation product 3a and the monoalkenylated cycloadduct was detected in  $16\%$  and  $25\%$  yields by  $^{1}H$ NMR, respectively (entries 8 and 9). The reaction efficiency was also sensitive to the reaction temperatures (entry 10). Some other rhodium complexes were also screened. For example, when  $[Cp*Rh(OAc)_2]$  was applied as a catalyst, the reaction still afforded dialkenylation product 3a, albeit in relatively lower yield (entry 11). However, the cationic rhodium species  $[Cp*Rh(MeCN)_3][SbF_6]$ , generated in situ by the addition of AgSbF<sub>6</sub>, totally inhibited the reaction (entry 12). A control experiment confirmed that the transformation did not proceed in the absence of the rhodium catalyst (entry 13). Finally,  $[\text{Ru}(p\text{-cymene})\text{Cl}_2]_2$  displayed no catalytic activity under the present reaction conditions (entry 14).

With the promising optimal conditions, we first investigated the influence of different substituents on the sulfonamide nitrogen atom (Scheme 2). In addition to the acetyl group, the butyryl group was also well suited and gave the product 3b

Table 1. Optimization of Oxidative Coupling between 1a and 2a<sup>a</sup>

				CO2Et	
	NHAc. 1a	2a	catalyst (3.0 mol %) oxidant (2.0 equiv) solvent (2.0 mL) 100°C, 24 h, Ar		O, N-Ac CO <sub>2</sub> Et 3a
entry	catalyst		oxidant (amt (equiv))	solvent	yield $(\%)^{b}$
1	$[Cp*RhCl_2]$		Cu(OAc) <sub>2</sub> ·H <sub>2</sub> O	toluene	99 (93) <sup>c</sup>
$\mathfrak{p}$	$[Cp*RhCl2]$ <sub>2</sub>		Cu(OAc) <sub>2</sub> ·H <sub>2</sub> O	<b>DME</b>	10
3	$[Cp*RhCl2]$		Cu(OAc), H, O	CH <sub>3</sub> CN	<b>NR</b>
$\overline{4}$	$[Cp*RhCl,$ ,	$Ag_2CO_3$		toluene	trace
5	$[Cp*RhCl2]$ <sub>2</sub>	AgOAc		toluene	48
6	$[Cp*RhCl_2],$	O,	$Cu(OAc)2·H2O (0.5) +$	toluene	trace
7	$[Cp*RhCl,$ ,		Cu(OAc) <sub>2</sub> ·H <sub>2</sub> O	toluene	82 <sup>d</sup>
8	$[Cp*RhCl,$ ,		Cu(OAc), H, O	toluene	$54^e$
9	$[Cp*RhCl_2],$		Cu(OAc) <sub>2</sub> ·H <sub>2</sub> O	toluene	$40^f$
10	$[Cp*RhCl2]$ <sub>2</sub>		Cu(OAc) <sub>2</sub> ·H <sub>2</sub> O	toluene	$NR^g$
11	$[Cp*Rh(OAc)2]$		Cu(OAc) <sub>2</sub> ·H <sub>2</sub> O	toluene	$65^h$
12	$[Cp*RhCl_2]$ AgSbF6		Cu(OAc) <sub>2</sub> ·H <sub>2</sub> O	toluene	NR <sup>i</sup>
13	none		Cu(OAc) <sub>2</sub> ·H <sub>2</sub> O	toluene	<b>NR</b>
14	$\lceil \text{Ru}(p\text{-cymene}) \rceil$ $\text{Cl}_2$ <sub>2</sub>		Cu(OAc) <sub>2</sub> ·H <sub>2</sub> O	toluene	<b>NR</b>

 ${}^a$ Reaction conditions: 1a (0.3 mmol), 2a (0.9 mmol).  ${}^b$ Yield determined by <sup>1</sup>H NMR (internal standard: 1,1,2,2-tetrachloroethane). NR = no reaction. <sup>c</sup> Value in parentheses indicates yield after purification.  $d$ 2.0 mol % of  $[Cp*RhCl<sub>2</sub>]$ <sub>2</sub> was used.  $e^d$ 1a:2a (1:1); the monoalkenylated cycloadduct was also formed in 16% yield. <sup>f</sup>1a:2a (2:1), 2a (0.3 mmol); the monoalkenylated cycloadduct was also  $f(x) = \ln 25\%$  yield.  ${}^{880}$  °C.  ${}^{h}$ 4.0 mol % of  $[Cp*Rh(OAc)_2]$  was used.<br>in 20 mol % of AgShE, was used.  $i$ 12.0 mol % of AgSbF<sub>6</sub> was used.

under the optimized reaction conditions in 97% yield. However, the pivaloyl, N-methoxy, and N-acrylyl derivatives were completely unreactive.

We next examined the scope of aryl sulfonamides with diverse arene substituents (Scheme 2). Gratifyingly, the reaction proceeded smoothly irrespective of the electronic nature of the substituents to afford th[e](#page-2-0) desired products in good to excellent yields. Many important functional groups, such as fluoro, chloro, bromo, methoxy, trifluoromethoxy, and ester substituents, were compatible in the present catalytic reaction (3d−i). Surprisingly, in addition to the desired product 3j, a substrate with a nitro substituent also afforded the nitro group reduced byproduct 3k in 16% yield. This result suggested that the nitro group might partially act as an oxidant in this reaction. The meta-substituted substrates were then studied. Both naphthalen-2-yl and 3-bromophenyl sulfonyl acetamide reacted with acrylates to form 3l−n as the sole products. Interestingly, the cyclizations occurred at the sterically more hindered position in compounds 3l−n, which were confirmed by NOESY spectroscopy of 31,m.<sup>17</sup> The exact reason for the selective ring formation is not clear yet, although it seems that the adjacent steric hindrance of me[ta-](#page-9-0)substituted group to the alkene could be partially involved. Notably, a substrate bearing a  $3$ -CF<sub>3</sub> moiety underwent the olefination to provide monoolefinic products 4a,b in moderate yield. The monofunctionalization in 4a,b might result from the combined electronic and steric effects of the meta-substituted  $CF_3$  group. Ortho-substituted aryl sulfonamides coupled with ethyl acrylate

<span id="page-2-0"></span>Scheme 2. Rhodium-Catalyzed Oxidative C−H Olefination with Acrylic Acid Esters<sup>a</sup>



 ${}^a$ Reaction on a 0.3 mmol scale. Yields of isolated products are given. NR = no reaction.  ${}^b$ The value in parentheses indicates the yield of the reaction on a 3.0 mmol scale.  $5.0$  mol % of  $[Cp*RhCl<sub>2</sub>]<sub>2</sub>$  was used.  $42.0$  equiv of 2 was used.  $e^{\text{At}}$  120  $^{\circ}\text{C}$ .

smoothly, leading to the formation of monoalkenylated cyclization products 4c−f. Moreover, various acrylates, including methyl acrylate, butyl acrylate, benzyl acrylate, and tert-butyl acrylate, efficiently reacted with 1a to produce the corresponding products 3o−r. In contrast to the acrylate, N,Ndimethylacrylamide reacted with 1a in a 1:1 ratio, resulting in the formation of cyclization product 4g. Nonetheless, other alkenes bearing electron-withdrawing groups, such as (phenylsulfonyl)ethane, diethyl vinylphosphonate, and ethyl vinyl ketone were not suitable substrates for this reaction and gave the desired products in very low yield  $(\leq 5\%)$ .

Subsequently, styrene was tested as a coupling partner in our catalytic reaction. However, no corresponding alkenylated product was observed under the optimized reaction conditions described above. Thus, the reaction was examined with various solvents and oxidants.<sup>17</sup> It is interesting to see in the present reaction system that copper acetate is advantageous in comparison to silver [a](#page-9-0)cetate as an oxidant in the reaction with acrylate esters, while the use of silver acetate gave a better result in the reaction with styrenes. Consequently, it was found that 1a coupled with styrene 5a (4.0 equiv) efficiently by employing AgOAc as oxidant and t-AmOH as solvent to give the dialkenylated derivative 6a in 81% yield (entry 1, Table 2). In this reaction, no cyclization process occurred. Under the reaction conditions, various substituted styrenes with eit[he](#page-3-0)r electron-donating or -withdrawing groups were good candidates and produced the corresponding products 6b−h in moderate to high yield (entries 2−8). The reaction is also tolerant of different substitutions on the aromatic sulfonamides. For example, 4-chloro- and 4-methoxy-substituted aryl sulfonamides also reacted well with 5a (entries 9 and 10). Examination of the meta-substituted substrate scope revealed that 3-bromophenyl sulfonyl acetamide only produced dialkenylated derivative 6k, albeit in low yield (entry 11). However, the reaction of a substrate bearing a  $3-CF_3$  group afforded monoalkenylated product 7a (entry 12), which is consistent with the result of the reaction with acrylate.

To our delight, the reactions can be easily performed on a large scale; for example, 3a and 6a were isolated in 86% (Scheme 2) and 70% yields (Table 2) on a 3.0 mmol scale, respectively. Since the N-acyl sulfonamide directing group is relatively labile, it can be easily re[mo](#page-3-0)ved under mildly basic conditions to give 8 (eq 1). The double bond of the enesulfonamide 8 and 6a can be reduced by NiCl<sub>2</sub>·6H<sub>2</sub>O/

## <span id="page-3-0"></span>Table 2. Rhodium-Catalyzed Oxidative C−H Olefination of 1 with Styrene $a$



a<br>Reaction on a 0.3 mmol scale. Yields of isolated products are given. b The value in parentheses indicates the yield of the reaction on a 3.0 mmol scale.

NaBH4 to give the saturated products 9 and 10, respectively (eq 2).



To probe the possible mechanism, we conducted some H/D exchange experiments with isotopically labeled solvents (Scheme 3a). When the reaction of 1a was conducted with  $D_2O$  in the absence of acrylate, 73% deuterium incorporation was observed at the two ortho positions of the aryl ring of 1a (Scheme 3a). This result suggested that a reversible cyclometalation mode was involved. If the same reaction was conducted in the presence of 2a, no deuterium incorporation was detected in recovered 1a, monoalkenylated product 3a′, and dialkenylated product 3a (Scheme 3b). This result is similar to those of the investigations by Loh and Zhang.<sup>18</sup> In both experiments, no N−H deuteration was observed, which could be rationalized by the easy N−D/H exchange durin[g t](#page-9-0)he workup on silica gel chromatography.

### ■ CONCLUSIONS

In conclusion, we have successfully established the Rh(III) catalyzed ortho C−H olefination of aryl sulfonamide directed by the  $SO<sub>2</sub>NHAc$  group. This oxidative coupling process is achieved highly efficiently and selectively with a broad substrate scope. The reactions of N-tosylacetamide with acrylate esters

Scheme 3. H/D Exchange Experiment



afford ortho-alkenylated benzofused five-membered cyclic sulfonamides, whereas styrenes provide the directed diolefination products. In Yu's report,  $SO_2NHC_6F_5$ -directed Pdcatalyzed benzenesulfonamide yielded exclusively the monoolefinated acyclic product. Thus, our method is complementary to previously reported methods based on the use of palladium catalyst.<sup>15d</sup> Considering the importance of aryl sulfonamides as well as aryl sultams as core structures in pharmacologically active s[ubs](#page-9-0)tances, this method should be attractive for both synthetic and medicinal chemistry.

#### **EXPERIMENTAL SECTION**

General Procedures. All reactions were carried out under an argon atmosphere using standard Schlenk techniques. <sup>1</sup>H NMR (400 MHz), <sup>19</sup>F NMR (376 MHz), and <sup>13</sup>C{<sup>1</sup>H} NMR (100 MHz) were recorded with  $CDCl<sub>3</sub>$  or  $DMSO-d<sub>6</sub>$  as solvent. Chemical shifts of <sup>1</sup>H, recorded with CDCl<sub>3</sub> or DMSO- $d_6$  as solvent. Chemical shifts of <sup>1</sup>H, <sup>19</sup>F, and <sup>13</sup>C{<sup>1</sup>H} NMR spectra are reported in parts per million (ppm). The residual solvent signals were used as references and the chemical shifts converted to the TMS scale (CDCl<sub>3</sub>,  $\delta_H$  7.26 ppm,  $\delta_C$ 77.00 ppm; DMSO- $d_6$ ,  $\delta_H$  2.50 ppm,  $\delta_C$  39.52 ppm). HRMS were measured on FTICR-mass spectrometers.  $[Cp*RhCl<sub>2</sub>]$ <sub>2</sub> was prepared from  $RhCl_3 \cdot xH_2O$  following a literature procedure.<sup>19</sup>

**General Procedure for Synthesis of Substrates 1.** General Procedure  $I^{20}$  The free sulfonamide  $(4.65 \text{ mmol}, 1.0 \text{ equiv})$  $(4.65 \text{ mmol}, 1.0 \text{ equiv})$  $(4.65 \text{ mmol}, 1.0 \text{ equiv})$ , anhydrous  $ZnCl_2$  (63.3 mg, 465  $\mu$ mol, 0.10 equiv), and Ac<sub>2</sub>O (3.95 mL, 41.8 mm[ol\)](#page-9-0) were mixed and stirred at 23 °C for 1−14 h and then poured into a mixture of EtOAc (50.0 mL) and  $H_2O$  (50.0 mL). The layers were separated, and the aqueous layer was extracted with EtOAc. The combined organic layers were dried over  $Na<sub>2</sub>SO<sub>4</sub>$ , concentrated in vacuo, and then purified by silica gel chromatography

(eluent EtOAc/petroleum ether 1/3, typically).<br>
General Procedure  $II^{27}$  To a cold suspension (0 °C) of 4methylbenzenesulfonamide (500 mg, 2.92 mmol) and powdered KOH (492 mg, 8.76 mmol, 3.0 [eq](#page-9-0)uiv) in  $CH_2Cl_2$  (10.0 mL) was added a solution of acyl chloride (2.92 mmol, 1.0 equiv) in  $CH_2Cl_2$  (10.0 mL). The mixture was stirred for 2 h at 23  $^{\circ}$ C, and water (20 mL) was added. After acidification with 1 N HCl solution (9.0 mL), the product was extracted with EtOAc  $(3 \times 50 \text{ mL})$ . The combined organic layers were dried over  $Na<sub>2</sub>SO<sub>4</sub>$ , concentrated in vacuo, and then purified by silica gel chromatography (eluent EtOAc/petroleum ether 1/3, typically).

General Procedure III. Methoxyamine hydrochloride (1003.3 mg, 12.0 mmol, 1.2 equiv) was added to a biphasic mixture of  $K_2CO_3$ (2763.3 mg, 20.0 mmol, 2.0 equiv) in a 2/1 mixture of EtOAc (72 mL) and H<sub>2</sub>O (36 mL). The resulting solution was cooled to 0  $^{\circ}$ C, followed by dropwise addition of 4-methylbenzene-1-sulfonyl chloride dissolved in a minimum amount of EtOAc. The reaction mixture was stirred overnight while it was warmed to room temperature. Afterward, the phases were separated and the aqueous phase was extracted twice with EtOAc. The combined organic layers were dried over  $MgSO_4$ , filtered, and evaporated under reduced pressure. The pure products were obtained without any further purification.

All phenylsulfonylacetamides were synthesized according to the literature procedures, and the yields were not optimized. Substrates 1a,b,d−g,i,j,l,m,p were synthesized through general procedure I. Substrates 1q−s were synthesized through general procedure II. For the known substrates,  $^1\mathrm{H}$  and  $^{13}\mathrm{C}$  NMR spectral data showed good agreement with the literature data.<sup>22</sup> Below we summarize characterization data for the newly synthesized phenyl sulfonylacetamides.

 $N-(4-Fluorophenylsulfonyl)acetamide (1c)$  $N-(4-Fluorophenylsulfonyl)acetamide (1c)$  $N-(4-Fluorophenylsulfonyl)acetamide (1c)$ . This compound was obtained as a white solid (807.6 mg, 80%) by following general procedure I: mp 124−126 °C; <sup>1</sup>H NMR (CDCl<sub>3</sub>, 400 MHz)  $\delta$  8.83 (s, 1H), 8.10 (dd, J = 8.7, 5.0 Hz, 2H), 7.23 (d, J = 8.5 Hz, 2H), 2.08 (s, 3H); <sup>13</sup>C{<sup>1</sup>H} NMR (CDCl<sub>3</sub>, 100 MHz)  $\delta$  168.3, 165.9 (d, J<sub>C−F</sub> = 257.2 Hz), 134.2 (d,  $J_{C-F}$  = 3.0 Hz), 131.3 (d,  $J_{C-F}$  = 9.7 Hz), 116.4 (d, J<sub>C−F</sub> = 22.8 Hz), 23.5; <sup>19</sup>F NMR (CDCl<sub>3</sub>, 376 MHz)  $\delta$  –102.4; IR (cm<sup>−</sup><sup>1</sup> ) ν 3121, 2901, 2806, 2694, 1694, 1586, 1458, 1350, 1229, 1097, 811, 537; HRMS (ESI) calcd for  $C_8H_9FNO_3S$   $[M + H]^+$  218.0282, found 218.0281.

Methyl 4-(N-Acetylsulfamoyl)benzoate (1h). This compound was obtained as a white solid (1051.9 mg, 88%) by following general procedure I: mp 188−190 °C; <sup>1</sup>H NMR (DMSO- $d_6$ , 400 MHz)  $\delta$ 12.31 (s, 1H), 8.17 (d, J = 7.9 Hz, 2H), 8.05 (d, J = 7.8 Hz, 2H), 3.90 (s, 3H), 1.94 (s, 3H); <sup>13</sup>C{<sup>1</sup>H} NMR (DMSO- $d_6$ , 100 MHz)  $\delta$  169.1, 165.1, 143.2, 133.9, 129.9, 128.0, 52.8, 23.3; IR (cm<sup>-1</sup>)  $\nu$  3224, 3108, 2951, 1735, 1565, 1420, 1333, 1300, 1163, 997, 860, 699, 595; HRMS (ESI) calcd for  $C_{10}H_{12}NO_5S$  [M + H]<sup>+</sup> 258.0431, found 258.0433.

N-(2-(Trifluoromethoxy)phenylsulfonyl)acetamide (1k). This compound was obtained as a white solid (1184.7 mg, 90%) by following general procedure I: mp 166−168 °C; <sup>1</sup>H NMR (DMSO- $d_6$ , 400 MHz)  $\delta$  12.53 (s, 1H), 8.07 (d, J = 7.7 Hz, 1H), 7.84 (t, J = 7.9 Hz, 1H), 7.61 (t, J = 6.9 Hz, 2H), 1.95 (s, 3H); <sup>13</sup>C{<sup>1</sup>H} NMR  $(DMSO-d<sub>6</sub>, 100 MHz)$  δ 168.9, 145.1, 136.1, 132.2, 131.1, 127.5, 120.9, 119.9 (q,  $J_{C-F}$  = 259.5 Hz), 23.0; <sup>19</sup>F NMR (DMSO- $d_6$ , 376 MHz)  $\delta$  -56.1; IR (cm<sup>-1</sup>)  $\nu$  3096, 2888, 2694, 1698, 1478, 1354, 1205, 1134, 1059, 856, 765, 603; HRMS (ESI) calcd for C<sub>9</sub>H<sub>9</sub>F<sub>3</sub>NO<sub>4</sub>S  $[M + H]$ <sup>+</sup> 284.0199, found 284.0205.

N-(3-Bromophenylsulfonyl)acetamide (1n). This compound was obtained as a white solid (956.6 mg, 74%) by following general procedure I: mp 131−133 °C; <sup>1</sup>H NMR (CDCl<sub>3</sub>, 400 MHz)  $\delta$  8.44 (s, 1H), 8.18 (s, 1H), 8.02 (d,  $J = 7.9$  Hz, 1H), 7.78 (d,  $J = 8.0$  Hz, 1H), 7.45 (t, J = 8.0 Hz, 1H), 2.10 (s, 3H); <sup>13</sup>C{<sup>1</sup>H} NMR (CDCl<sub>3</sub>, 100 MHz) δ 168.3, 140.0, 137.2, 131.0, 130.6, 127.0, 122.8, 23.6; IR (cm<sup>−</sup><sup>1</sup> ) ν 3088, 2880, 2806, 1702, 1474, 1354, 1155, 1006, 790, 661, 558; HRMS (ESI) calcd for  $C_8H_9BrNO_3S [M + H]^+$  277.9481, found 277.9483.

N-(3-(Trifluoromethyl)phenylsulfonyl)acetamide (1o). This compound was obtained as a white solid (968.4 mg, 78%) by following general procedure I: mp 125−127 °C;  $^1\text{H NMR}$  (CDCl $_3$ , 400 MHz)  $\delta$ 8.65 (s, 1H), 8.31 (d, J = 5.7 Hz, 2H), 7.93 (d, J = 7.8 Hz, 1H), 7.73 (t,  $J = 8.2$  Hz, 1H), 2.11 (s, 3H); <sup>13</sup>C{<sup>1</sup>H} NMR (CDCl<sub>3</sub>, 100 MHz)  $\delta$ 168.4, 139.4, 131.8, 131.7 (d,  $J_{C-F}$  = 33.3 Hz), 130.8 (d,  $J_{C-F}$  = 3.3 Hz), 129.9, 125.3 (q, J<sub>C−F</sub> = 3.6 Hz), 123.0 (d, J<sub>C−F</sub> = 273.0 Hz), 23.5;<br><sup>19</sup>F NMR (CDCl<sub>3</sub>, 376 MHz) δ −62.8; IR (cm<sup>−1</sup>) ν 3092, 2888, 2718, 1690, 1495, 1329, 1171, 1118, 1010, 798, 603; HRMS (ESI) calcd for  $C_9H_9F_3NO_3S$  [M + H]<sup>+</sup> 268.0250, found 268.0250.

N-Tosylacrylamide (1s). This compound was obtained as a white solid (499.3 mg, 76%) by following general procedure II: mp 158−160  $^{\circ}$ C; <sup>1</sup>H NMR (CDCl<sub>3</sub>, 400 MHz)  $\delta$  8.49 (bs, 1H), 7.98 (d, J = 8.1 Hz, 2H), 7.35 (d, J = 8.2 Hz, 2H), 6.38 (d, J = 17.0 Hz, 1H), 6.10 (dd, J = 16.7, 10.7 Hz, 1H), 5.83 (d, J = 10.5 Hz, 1H), 2.44 (s, 3H); <sup>13</sup>C{<sup>1</sup>H} NMR (CDCl<sub>3</sub>, 100 MHz)  $\delta$  162.7, 145.3, 135.3, 131.5, 129.7, 129.6, 128.4, 21.7; IR (cm<sup>-1</sup>)  $\nu$  3622, 3228, 2855, 1702, 1599, 1441, 1325,

1155, 1076, 819, 549; HRMS (ESI) calcd for  $C_{10}H_{12}NO_3S$   $[M + H]^+$ 226.0532, found 226.0533.

N-Methoxy-4-methylbenzenesulfonamide (1t). This compound was obtained as a white solid (2170.8 mg, 90%) by following general procedure III: mp 109−111 °C; <sup>1</sup>H NMR (CDCl<sub>3</sub>, 400 MHz)  $\delta$  7.81  $(d, J = 8.3 \text{ Hz}, 2H), 7.35 \ (d, J = 8.1 \text{ Hz}, 2H), 7.01 \ (s, 1H), 3.79 \ (s,$ 3H), 2.45 (s, 3H); <sup>13</sup>C{<sup>1</sup>H} NMR (CDCl<sub>3</sub>, 100 MHz) 144.9, 133.5, 129.7, 128.5, 65.0, 21.7; IR (cm<sup>-1</sup>)  $\nu$  3224, 2938, 1922, 1590, 1400, 1329, 1159, 1030, 922, 827, 748, 570; HRMS (ESI) calcd for  $C_8H_{12}NO_3S$  [M + H]<sup>+</sup> 202.0532, found 202.0537.

Preparation of the Product 3 or 4 (General Procedure A). A mixture of sulfonamide 1 (0.30 mmol, 1.0 equiv),  $[Cp*RhCl_2]$ <sub>2</sub> (5.6) mg, 0.009 mmol, 3.0 mol %), and  $Cu(OAc)<sub>2</sub>·H<sub>2</sub>O$  (120 mg, 0.60 mmol, 2.0 equiv) were weighed in a 50 mL Schlenk tube equipped with a stir bar. Dry toluene (2.0 mL) was added followed immediately by the alkene 2 (0.9 mmol, 3.0 equiv), and the mixture was stirred at 100 °C for 24 h under an Ar atmosphere. Afterward, the vial was cooled to room temperature, diluted with  $CH_2Cl_2$ , and transferred to a round-bottom flask. Silica was added to the flask, and the volatiles were evaporated under reduced pressure. The purification was performed by flash column chromatography on silica gel (eluent EtOAc/petroleum ether 1/3, typically).

Ethyl (N-Acetyl-1,1-dioxo-5-methyl-(E)-7-(ethoxy-3-oxoprop-1 enyl)benzo[d]isothiazol-3-yl)acetate (3a). This compound was obtained as a light yellow solid (115.0 mg, 93%) by following general procedure A: mp 126−128 °C; <sup>1</sup>H NMR (CDCl<sub>3</sub>, 400 MHz)  $\delta$  8.02  $(d, J = 15.9 \text{ Hz}, 1\text{H}), 7.55 \text{ (s, 1H)}, 7.34 \text{ (s, 1H)}, 6.63 \text{ (d, } J = 15.9 \text{ Hz},$ 1H), 5.65 (d, J = 6.2 Hz, 1H), 4.29 (q, J = 7.1 Hz, 2H), 4.14 (q, J = 7.1 Hz, 2H), 3.16−3.10 (m, 1H), 2.93 (dd, J = 16.1, 7.7 Hz, 1H), 2.61 (s, 3H), 2.48 (s, 3H), 1.34 (t, J = 7.1 Hz, 3H), 1.20 (t, J = 7.1 Hz, 3H);<br><sup>13</sup>C{<sup>1</sup>H} NMR (CDCl<sub>3</sub>, 100 MHz)  $\delta$  169.5, 167.3, 165.4, 145.6, 136.1, 135.6, 130.8, 129.4, 128.7, 126.0, 124.4, 61.1, 61.0, 54.7, 39.0, 23.6, 21.8, 14.2, 14.0; IR (cm<sup>-1</sup>) ν 3560, 3411, 2984, 1735, 1706, 1378, 1275, 1212, 1026, 868, 619; HRMS (ESI) calcd for  $C_{19}H_{24}NO_7S$  [M + H]<sup>+</sup> 410.1268, found 410.1275.

Ethyl (N-Butyryl-1,1-dioxo-5-methyl-(E)-7-(ethoxy-3-oxoprop-1 enyl)benzo[d]isothiazol-3-yl)acetate (3b). This compound was obtained as a white solid (128.7 mg, 97%) by following general procedure A: mp 96−98 °C; <sup>1</sup>H NMR (CDCl<sub>3</sub>, 400 MHz)  $\delta$  7.97 (d, J  $= 15.9$  Hz, 1H), 7.52 (s, 1H), 7.31 (s, 1H), 6.58 (d, J = 15.9 Hz, 1H), 5.62 (dd, J = 7.5, 3.2 Hz, 1H), 4.24 (q, J = 7.1 Hz, 2H), 4.08 (t, J = 7.1 Hz, 2H), 3.06 (dd,  $J = 16.0$ , 3.3 Hz, 1H), 2.94 (dd,  $J = 16.2$ , 7.5 Hz, 1H), 2.91−2.79 (m, 2H), 2.44 (s, 3H), 1.80−1.71 (m, 2H), 1.30 (t, J = 7.1 Hz, 3H), 1.15 (t, J = 7.1 Hz, 3H), 0.98 (t, J = 7.4 Hz, 3H);<br><sup>13</sup>C{<sup>1</sup>H} NMR (CDCl<sub>3</sub>, 100 MHz)  $\delta$  170.5, 169.4, 165.3, 145.4, 136.1, 135.5, 130.7, 129.6, 128.6, 125.9, 124.3, 61.0, 60.9, 54.6, 39.0, 37.5, 21.7, 17.7, 14.1, 13.9, 13.4; IR (cm<sup>−1</sup>)  $\nu$  3394, 2979, 1731, 1640, 1320, 1179, 1071, 972, 669, 540; HRMS (ESI) calcd for  $C_{21}H_{28}NO_7S$  [M + H]<sup>+</sup> 438.1581, found 438.1587.

Ethyl (N-Acetyl-1,1-dioxo-(E)-7-(ethoxy-3-oxoprop-1-enyl)benzo- [d]isothiazol-3-yl) acetate  $(3c)$ . This compound was obtained as a light yellow solid (106.8 mg, 90%) by following general procedure A: mp 89−91 °C; <sup>1</sup>H NMR (CDCl<sub>3</sub>, 400 MHz)  $\delta$  8.07 (d, J = 15.9 Hz, 1H), 7.76 (d, J = 7.7 Hz, 1H), 7.69 (t, J = 7.7 Hz, 1H), 7.57 (d, J = 7.7 Hz, 1H), 6.64 (d, J = 15.9 Hz, 1H), 5.71 (dd, J = 7.7, 3.3 Hz, 1H), 4.30  $(q, J = 7.1 \text{ Hz}, 2\text{H})$ , 4.13  $(q, J = 7.1 \text{ Hz}, 2\text{H})$ , 3.16  $(dd, J = 16.1, 3.3$ Hz, 1H), 2.95 (dd,  $J = 16.1, 7.8$  Hz, 1H), 2.62 (s, 3H), 1.35 (t,  $J = 7.1$ Hz, 3H), 1.19 (t, J = 7.1 Hz, 3H); <sup>13</sup>C{<sup>1</sup>H} NMR (CDCl<sub>3</sub>, 100 MHz) δ 169.5, 167.3, 165.3, 135.9, 135.4, 134.3, 132.1, 131.2, 127.8, 125.7, 124.8, 61.2, 61.1, 54.8, 39.0, 23.6, 14.2, 14.0; IR (cm<sup>−</sup><sup>1</sup> ) 3564, 3481, 2992, 1731, 1702, 1374, 1283, 1158, 1042, 959, 594; HRMS (ESI) calcd for  $C_{18}H_{22}NO_7S$   $[M + H]^+$  396.1112, found 396.1119.

Ethyl (N-Acetyl-1,1-dioxo-5-fluoro-(E)-7-(ethoxy-3-oxoprop-1 enyl)benzo[d]isothiazol-3-yl)acetate (3d). This compound was obtained as a light yellow solid (116.1 mg, 94%) by following general procedure A: mp 141−143 °C; <sup>1</sup>H NMR (CDCl<sub>3</sub>, 400 MHz)  $\delta$  7.97  $(d, J = 15.8 \text{ Hz}, 1\text{H}), 7.42 (d, J = 9.0 \text{ Hz}, 1\text{H}), 7.30 (t, J = 6.7 \text{ Hz}, 1\text{H}),$ 6.60 (d, J = 15.8 Hz, 1H), 5.66 (dd, J = 8.0, 2.7 Hz, 1H), 4.27 (q, J = 7.1 Hz, 2H), 4.13 (q, J = 7.1 Hz, 2H), 3.15 (dd, J = 16.4, 3.0 Hz, 1H), 2.91 (dd, J = 16.3, 8.1 Hz, 1H), 2.58 (s, 3H), 1.32 (t, J = 7.1 Hz, 3H),

1.20 (d, J = 7.1 Hz, 3H); <sup>13</sup>C{<sup>1</sup>H} NMR (CDCl<sub>3</sub>, 100 MHz)  $\delta$  169.4, 167.0, 165.7 (d,  $J_{C-F} = 257.1$  Hz), 164.9, 139.1 (d,  $J_{C-F} = 10.4$  Hz), 134.3, 134.0 (d,  $J_{C-F}$  = 9.0 Hz), 128.2 (d,  $J_{C-F}$  = 2.0 Hz), 126.0, 115.3 (d, J<sub>C−F</sub> = 24.9 Hz), 113.1 (d, J<sub>C−F</sub> = 25.1 Hz), 61.3, 61.2, 54.5 (d, J<sub>C−F</sub> = 2.0 Hz), 38.7, 23.6, 14.1, 13.9; <sup>19</sup>F NMR (CDCl<sub>3</sub>, 376 MHz)  $\delta$ −101.3; IR (cm<sup>−</sup><sup>1</sup> ) ν 3407, 3071, 2988, 1731, 1710, 1382, 1270, 1158, 1038, 959, 860, 632; HRMS (ESI) calcd for  $C_{18}H_{24}FN_{2}O_{7}S$  [M + NH4] <sup>+</sup> 431.1283, found 431.1285.

Ethyl (N-Acetyl-1,1-dioxo-5-chloro-(E)-7-(ethoxy-3-oxoprop-1 enyl)benzo[d]isothiazol-3-yl)acetate (3e). This compound was obtained as a light yellow solid (124.3 mg, 96%) by following general procedure A: mp 124−126 °C; <sup>1</sup>H NMR (CDCl<sub>3</sub>, 400 MHz)  $\delta$  7.94  $(d, J = 15.8 \text{ Hz}, 1\text{H}), 7.70 \text{ (s, 1H)}, 7.56 \text{ (s, 1H)}, 6.62 \text{ (d, } J = 15.9 \text{ Hz},$ 1H), 5.65 (d, J = 7.5 Hz, 1H), 4.27 (q, J = 7.1 Hz, 2H), 4.12 (q, J = 7.1 Hz 2H), 3.13 (d,  $J = 16.4$  Hz, 1H), 2.92 (dd,  $J = 16.3$ , 8.0 Hz, 1H), 2.58 (s, 3H), 1.32 (t,  $J = 7.0$  Hz, 3H), 1.19 (t,  $J = 6.7$  Hz, 3H); 2.58 (s, 3H), 1.32 (t, J = 7.0 Hz, 3H), 1.19 (t, J = 6.7 Hz, 3H);  $^{13}C(^{1}H)$  NMR (CDCl<sub>3</sub>, 100 MHz)  $\delta$  169.3, 167.0, 164.9, 140.8, 137.7, 134.1, 132.6, 130.5, 127.9, 125.9, 125.7, 61.3, 61.2, 54.4, 38.6, 23.6, 14.1, 13.9; IR (cm<sup>-1</sup>) ν 3075, 2992, 1731, 1702, 1569, 1432, 1266, 1171, 1021, 955, 856, 615; HRMS (ESI) calcd for  $C_{18}H_{21}CINO_{7}S$  [M  $+ H$ <sup>+</sup> 430.0722, found 430.0719.

Ethyl (N-Acetyl-1,1-dioxo-5-bromo-(E)-7-(ethoxy-3-oxoprop-1 enyl)benzo[d]isothiazol-3-yl)acetate (3f). This compound was obtained as a yellow solid (118.7 mg, 84%) by following general procedure A: mp 129−130 °C; <sup>1</sup>H NMR (CDCl<sub>3</sub>, 400 MHz)  $\delta$  7.96  $(d, J = 15.9 \text{ Hz}, 1\text{H})$ , 7.87 (s, 1H), 7.73 (s, 1H), 6.63 (d, J = 15.9 Hz, 1H), 5.66 (dd, J = 8.0, 3.1 Hz, 1H), 4.28 (q, J = 7.1 Hz, 2H), 4.15 (q, J  $= 7.1$  Hz, 2H), 3.14 (dd, J = 16.4, 3.2 Hz, 1H), 2.93 (dd, J = 16.4, 8.0) Hz, 1H), 2.59 (s, 3H), 1.33 (t,  $J = 7.1$  Hz, 3H), 1.21 (t,  $J = 7.1$  Hz, 3H); <sup>13</sup>C{<sup>1</sup>H} NMR (CDCl<sub>3</sub>, 100 MHz)  $\delta$  169.3, 167.0, 164.9, 137.7, 134.1, 132.6, 131.0, 130.9, 129.1, 128.7, 126.0, 61.3, 61.2, 54.4, 38.7, 23.6, 14.2, 14.0; IR (cm<sup>-1</sup>) ν 3075, 2984, 1714, 1565, 1316, 1279, 1167, 1042, 968, 619, 569; HRMS (ESI) calcd for  $C_{18}H_{21}BrNO_7S$  [M + H]+ 474.0217, found 474.0220.

Ethyl (N-Acetyl-1,1-dioxo-5-methoxy-(E)-7-(ethoxy-3-oxoprop-1 enyl)benzo[d]isothiazol-3-yl)acetate  $(3g)$ . This compound was obtained as a white solid (91.0 mg, 71%) by following general procedure A: mp 147−149 °C; <sup>1</sup>H NMR (CDCl<sub>3</sub>, 400 MHz)  $\delta$  7.97  $(d, J = 15.9 \text{ Hz}, 1\text{H}), 7.19 \text{ (d, } J = 1.9 \text{ Hz}, 1\text{H}), 7.01 \text{ (d, } J = 1.7 \text{ Hz},$ 1H), 6.60 (d, J = 15.9 Hz, 1H), 5.63 (dd, J = 8.0, 3.2 Hz, 1H), 4.27 (q,  $J = 7.1$  2H), 4.13 (q,  $J = 7.1$  2H), 3.88 (s, 3H), 3.14 (dd,  $J = 16.2$ , 3.3 Hz, 1H), 2.89 (dd,  $J = 16.2$ , 8.1 Hz, 1H), 2.59 (s, 3H), 1.33 (t,  $J = 7.1$ Hz, 3H), 1.20 (t, J = 7.2 Hz, 3H); <sup>13</sup>C{<sup>1</sup>H} NMR (CDCl<sub>3</sub>, 100 MHz) δ 169.7, 167.2, 165.3, 164.1, 138.4, 135.5, 132.7, 124.8, 124.1, 114.4, 109.8, 61.2, 61.0, 56.0, 54.6, 39.1, 23.5, 14.2, 14.0; IR (cm<sup>-1</sup>) ν 3087, 2979, 2905, 1727, 1710, 1469, 1316, 1287, 1150, 1026, 968, 856, 636; HRMS (ESI) calcd for  $C_{19}H_{24}NO_8S$  [M + H]<sup>+</sup> 426.1217, found 426.1223.

Ethyl (N-Acetyl-1,1-dioxo-5-trifluoromethoxy-(E)-7-(ethoxy-3-oxoprop-1-enyl)benzo[d]isothiazol-3-yl)acetate (3h). This compound was obtained as a yellow solid (115.5 mg, 80%) by following general procedure A: mp 101−103 °C; <sup>1</sup>H NMR (CDCl<sub>3</sub>, 400 MHz)  $\delta$  8.01  $(d, J = 15.9 \text{ Hz}, 1\text{H}), 7.54 \text{ (s, 1H)}, 7.46 \text{ (s, 1H)}, 6.65 \text{ (d, } J = 15.9 \text{ Hz},$ 1H), 5.72 (dd, J = 8.4, 3.0 Hz, 1H), 4.30 (q, J = 7.1 Hz, 2H), 4.15 (q, J  $= 7.1$  Hz, 2H), 3.20 (dd, J = 16.4, 3.2 Hz, 1H), 2.90 (dd, J = 16.4, 8.4 Hz, 1H), 2.61 (s, 3H), 1.35 (t,  $J = 7.1$  Hz, 3H), 1.21 (t,  $J = 7.1$  Hz, 3H); <sup>13</sup>C{<sup>1</sup>H} NMR (CDCl<sub>3</sub>, 100 MHz)  $\delta$  169.4, 167.0, 164.9, 153.2  $(q, J<sub>C-F</sub> = 1.9 Hz)$ , 138.7, 134.1, 133.6, 130.3, 126.3, 120.0 (d, J<sub>C−F</sub> = 260.8 Hz), 119.9, 117.5 (d, J<sub>C−F</sub> = 1.0 Hz), 61.4, 61.3, 54.6, 38.8, 23.6, 14.2, 13.9; <sup>19</sup>F NMR (CDCl<sub>3</sub>, 376 MHz)  $\delta$  –57.7; IR (cm<sup>-1</sup>)  $\nu$  3407, 3083, 2996, 1710, 1590, 1366, 1258, 1171, 1038, 972, 851, 694; HRMS (ESI) calcd for  $C_{19}H_{21}F_3NO_8S$  [M + H]<sup>+</sup> 480.0935, found 480.0940.

Ethyl (N-Acetyl-1,1-dioxo-5-methoxycarbonyl-(E)-7-(ethoxy-3-oxoprop-1-enyl)benzo[d]isothiazol-3-yl) acetate (3i). This compound was obtained as a white solid (101.0 mg, 74%) by following general procedure A: mp 138−140 °C; <sup>1</sup>H NMR (CDCl<sub>3</sub>, 400 MHz)  $\delta$  8.41  $(s, 1H)$ , 8.17  $(s, 1H)$ , 8.06  $(d, J = 15.9 \text{ Hz}, 1H)$ , 6.74  $(d, J = 15.9 \text{ Hz},$ 1H), 5.72 (dd, J = 7.3, 3.3 Hz, 1H), 4.31 (q, J = 7.1 Hz, 2H), 4.13 (q, J  $= 7.1$  Hz, 2H), 3.99 (s, 3H), 3.16 (dd, J = 16.1, 3.4 Hz, 1H), 3.03 (dd,

 $J = 16.1, 7.4$  Hz, 1H), 2.63 (s, 3H), 1.35 (t,  $J = 7.1$  Hz, 3H), 1.19 (t,  $J =$ 7.1 Hz, 3H); <sup>13</sup>C{<sup>1</sup>H} NMR (CDCl<sub>3</sub>, 100 MHz)  $\delta$  169.0, 167.1, 165.0, 164.4, 136.4, 135.7, 135.2, 134.4, 131.5, 128.7, 126.4, 125.7, 61.2, 61.1, 54.8, 53.0, 38.6, 23.6, 14.1, 13.9; IR (cm<sup>-1</sup>)  $\nu$  3398, 3079, 2992, 1723, 1706, 1436, 1362, 1270, 1158, 968, 764, 607; HRMS (ESI) calcd for  $C_{20}H_{24}NO_9S$  [M + H]<sup>+</sup> 454.1166, found 454.1174.

Ethyl (N-Acetyl-1,1-dioxo-5-nitro-(E)-7-(ethoxy-3-oxoprop-1-enyl) benzo[d]isothiazol-3-yl)acetate (3j). This compound was obtained as a light yellow solid (72.2 mg, 55%) by following general procedure A: mp 105−107 °C; <sup>1</sup> H NMR (CDCl3, 400 MHz) δ 8.57 (s, 1H), 8.41 (s, 1H), 8.02 (d, J = 15.8 Hz, 1H), 6.78 (d, J = 15.8 Hz, 1H), 5.77 (d, J = 7.4 Hz, 1H), 4.30 (q,  $J = 6.9$  Hz, 2H), 4.13 (q,  $J = 7.0$  Hz, 2H), 3.20  $(d, J = 16.5 Hz, 1H), 3.02 (dd, J = 16.4, 7.8 Hz, 1H), 2.60 (s, 3H), 1.34$  $(t, J = 7.0 \text{ Hz}, 3\text{H})$ , 1.20  $(t, J = 7.1 \text{ Hz}, 3\text{H})$ ; <sup>13</sup>C{<sup>1</sup>H} NMR (CDCl<sub>3</sub>, 100 MHz) δ 169.0, 166.8, 164.6, 151.2, 138.1, 136.5, 133.2, 133.0, 127.2, 122.6, 120.6, 61.43, 61.36, 54.7, 38.2, 23.6, 14.1, 13.9; IR (cm<sup>−1</sup>) ν 3419, 3091, 2971, 1723, 1552, 1328, 1275, 1167, 1026, 963, 744, 611; HRMS (ESI) calcd for  $C_{18}H_{21}N_2O_9S$  [M + H]<sup>+</sup> 441.0962, found 441.0966.

Ethyl (1,1-Dioxo-2-acetyl-5-amino-(E)-7-(ethoxy-3-oxoprop-1 enyl)-benzo[d]isothiazol-3-yl)acetate (3k). This compound was obtained as a brown solid (19.5 mg, 16%) by following general procedure A: mp 157−159 °C; <sup>1</sup> H NMR (CDCl3, 400 MHz) δ 7.91  $(d, J = 15.9 \text{ Hz}, 1H)$ , 6.87 (s, 1H), 6.68 (s, 1H), 6.53 (d, J = 15.9 Hz, 1H), 5.55 (d, J = 7.6 Hz, 1H), 4.44 (s, 2H), 4.27 (q, J = 7.1 Hz, 2H), 4.14 (q,  $J = 7.1$  Hz, 2H), 3.09 (d,  $J = 16.2$  Hz, 1H), 2.88 (dd,  $J = 16.0$ , 7.9 Hz, 1H), 2.58 (s, 3H), 1.33 (t,  $J = 7.1$  Hz, 3H), 1.21 (t,  $J = 7.0$  Hz, 3H); <sup>13</sup>C{<sup>1</sup>H} NMR (CDCl<sub>3</sub>, 100 MHz)  $\delta$  169.9, 167.4, 165.6, 152.0, 138.3, 136.2, 132.4, 124.1, 120.6, 113.5, 109.2, 61.1, 61.0, 54.5, 39.2, 23.5, 14.2, 14.0; IR (cm<sup>-1</sup>) *ν* 3469, 3378, 3253, 2975, 2913, 1743, 1706, 1631, 1374, 1304, 1196, 1042, 963, 681; HRMS (ESI) calcd for  $C_{18}H_{23}N_2O_7S$   $[M + H]^+$  411.1222, found 411.1222.

Ethyl (N-Acetyl-1,1-dioxo-(E)-9-(ethoxy-3-oxoprop-1-enyl) naphtho[1,2-d]isothiazol-3-yl)acetate (3l). This compound was obtained as a white solid (54.4 mg, 41%) by following general procedure A: mp 143−145 °C; <sup>1</sup>H NMR (CDCl<sub>3</sub>, 400 MHz)  $\delta$  8.19  $(S, 1H)$ , 8.11 (d, J = 15.9 Hz, 1H), 8.08–8.04 (m, 1H), 8.03–7.98 (m, 1H), 7.74 (dd, J = 6.1, 3.1 Hz, 2H), 6.72 (d, J = 15.8 Hz, 1H), 6.16 (t,  $J = 3.3$  Hz, 1H), 4.29 (q,  $J = 7.1$  Hz, 2H), 4.13 (dd,  $J = 6.9$ , 5.2 Hz, 2H), 3.21 (dd,  $J = 15.6$ , 3.2 Hz, 1H), 2.93 (dd,  $J = 15.6$ , 5.2 Hz, 1H), 2.65 (s, 3H), 1.35 (t,  $J = 7.1$  Hz, 3H), 1.18 (t,  $J = 7.1$  Hz, 3H); 2.65 (s, 3H), 1.35 (t, J = 7.1 Hz, 3H), 1.18 (t, J = 7.1 Hz, 3H);  $^{13}C(^{1}H)$  NMR (CDCl<sub>3</sub>, 100 MHz)  $\delta$  169.5, 167.1, 165.5, 136.0, 135.3, 133.9, 129.98, 129.96, 129.7, 129.54, 129.47, 126.9, 125.9, 123.8, 123.7, 61.3, 60.9, 54.6, 40.5, 23.5, 14.2, 13.9; IR (cm<sup>-1</sup>) ν 2918, 1740, 1694, 1329, 1288, 1159, 1035, 757, 512; HRMS (ESI) calcd for  $C_{22}H_{24}NO_7S$   $[M + H]^+$  446.1268, found 446.1271.

Ethyl (N-Acetyl-1,1-dioxo-4-bromo-(E)-7-(ethoxy-3-oxoprop-1 enyl)benzo[d]isothiazol-3-yl)acetate (3m). This compound was obtained as a yellow solid (116.0 mg, 82%) by following general procedure A: mp 117−119 °C; <sup>1</sup>H NMR (CDCl<sub>3</sub>, 400 MHz)  $\delta$  8.02  $(d, J = 15.9 \text{ Hz}, 1\text{H})$ , 7.85  $(d, J = 8.3 \text{ Hz}, 1\text{H})$ , 7.65  $(d, J = 8.3 \text{ Hz},$ 1H), 6.63 (d,  $J = 15.9$  Hz, 1H), 5.64 (t,  $J = 3.4$  Hz, 1H), 4.29 (q,  $J =$ 7.1 Hz, 2H), 4.03 (q,  $J = 7.1$  Hz, 2H), 3.31 (dd,  $J = 15.3$ , 3.9 Hz, 1H), 3.24 (dd, J = 15.4, 4.3 Hz, 1H), 2.62 (s, 3H), 1.34 (t, J = 7.1 Hz, 3H), 1.10 (t,  $J = 7.1$  Hz, 3H); <sup>13</sup>C{<sup>1</sup>H} NMR (CDCl<sub>3</sub>, 100 MHz)  $\delta$  168.6, 167.0, 165.2, 137.9, 134.7, 134.5, 130.3, 129.0, 125.2, 120.2, 61.19, 61.15, 56.2, 35.8, 23.7, 14.2, 13.8 (one signal missing due to overlap); IR (cm<sup>−</sup><sup>1</sup> ) ν 2988, 1723, 1648, 1466, 1263, 1117, 1026, 819, 603, 537; HRMS (ESI) calcd for  $C_{18}H_{21}BrNO_7S$  [M + H]<sup>+</sup> 474.0217, found 474.0215.

Butyl (N-Acetyl-1,1-dioxo-4-bromo-(E)-7-(butoxy-3-oxoprop-1 enyl)benzo[d]isothiazol-3-yl)acetate  $(3n)$ . This compound was obtained as a colorless oil (103.2 mg, 60%) by following general procedure A: <sup>1</sup>H NMR (CDCl<sub>3</sub>, 400 MHz)  $\delta$  8.02 (d, J = 15.9 Hz, 1H), 7.85 (d, J = 8.3 Hz, 1H), 7.65 (d, J = 8.3 Hz, 1H), 6.64 (d, J = 15.9 Hz, 1H), 5.64 (t, J = 4.1 Hz, 1H), 4.23 (t, J = 6.7 Hz, 2H), 3.98 (t,  $J = 6.8$  Hz, 2H), 3.32 (dd,  $J = 15.3$ , 4.0 Hz, 1H), 3.26 (dd,  $J = 15.4$ , 4.3 Hz, 1H), 2.62 (s, 3H), 1.73−1.66 (m, 2H), 1.48−1.40 (m, 4H), 1.23−<br>1.20 (m, 2H), 0.96 (t, J = 7.4 Hz, 3H), 0.86 (t, J = 7.4 Hz, 3H); 1.20 (m, 2H), 0.96 (t, J = 7.4 Hz, 3H), 0.86 (t, J = 7.4 Hz, 3H);  $^{13}C(^{1}H)$  NMR (CDCl<sub>3</sub>, 100 MHz)  $\delta$  168.6, 167.0, 165.2, 137.9, 134.7,

134.58, 134.52, 130.4, 128.9, 125.1, 120.1, 65.09, 65.05, 56.2, 35.8, 30.6, 30.2, 23.6, 19.1, 18.9, 13.68, 13.62; IR (cm<sup>-1</sup>) ν 2955, 2864, 1711, 1466, 1321, 1254, 1155, 819; HRMS (ESI) calcd for  $C_{22}H_{29}BrNO_7S$  [M + H]<sup>+</sup> 530.0843, found 530.0849.

Ethyl (N-Acetyl-1,1-dioxo-6-trifluoromethylbenzo[d]isothiazol-3 yl)acetate (4a). This compound was obtained as a white solid (64.2 mg, 59%) by following general procedure A except using 5.0 mol % of  $[Cp*RhCl<sub>2</sub>]$ <sub>2</sub> at 120 °C: mp 94–95 °C; <sup>1</sup>H NMR (CDCl<sub>3</sub>, 400 MHz)  $\delta$  8.09 (s, 1H), 7.94 (d, J = 8.2 Hz, 1H), 7.79 (d, J = 8.2 Hz, 1H), 5.77  $(d, J = 7.7 \text{ Hz}, 1H), 4.12 (q, J = 7.1 \text{ Hz}, 2H), 3.20 (dd, J = 16.4, 1.5$ Hz, 1H), 2.95 (dd,  $J = 16.4$ , 8.2 Hz, 1H), 2.60 (s, 3H), 1.19 (t,  $J = 7.1$ Hz, 3H); <sup>13</sup>C{<sup>1</sup>H} NMR (CDCl<sub>3</sub>, 100 MHz)  $\delta$  169.3, 167.1, 138.7 (d,  $J_{C-F}$  = 0.8 Hz), 134.7, 132.8 (q,  $J_{C-F}$  = 34.0 Hz), 131.1 (q,  $J_{C-F}$  = 3.4 Hz), 126.1, 122.7 (q,  $J_{C-F} = 273.1$  Hz), 119.5 (q,  $J_{C-F} = 4.0$  Hz), 61.3, 55.3, 38.6, 23.6, 14.0; <sup>19</sup>F NMR (CDCl<sub>3</sub>, 376 MHz)  $\delta$  –62.8; IR (cm<sup>−</sup><sup>1</sup> ) ν 3075, 3000, 2946, 2905, 1735, 1714, 1627, 1416, 1312, 1133, 1034, 864, 607, 528; HRMS (ESI) calcd for  $C_{14}H_{15}F_3NO_5S$   $[M + H]^+$ 366.0618, found 366.0623.

Butyl (N-Acetyl-1,1-dioxo-6-trifluoromethylbenzo[d]isothiazol-3 yl) acetate (4b). This compound was obtained as a yellow oil (50.7 mg, 43%) by following general procedure A except using 5.0 mol % of  $[Cp*RhCl<sub>2</sub>]<sub>2</sub>$  at 120 °C: <sup>1</sup>H NMR (CDCl<sub>3</sub>, 400 MHz)  $\delta$  8.09 (s, 1H), 7.94 (d,  $J = 8.2$  Hz, 1H), 7.79 (d,  $J = 8.4$  Hz, 1H), 5.78 (dd,  $J = 8.2$ , 3.0 Hz, 1H), 4.08 (t, J = 6.7 Hz, 2H), 3.22 (dd, J = 16.4, 3.2 Hz, 1H), 2.97 (dd, J = 16.4, 8.2 Hz, 1H), 2.61 (s, 3H), 1.55−1.50 (m, 2H), 1.30− 1.24 (m, 2H), 0.88 (t, J = 7.4 Hz, 3H); <sup>13</sup>C{<sup>1</sup>H} NMR (CDCl<sub>3</sub>, 100 MHz)  $\delta$  169.4, 167.2, 138.7, 134.7, 132.9 (q, J<sub>C−F</sub> = 34.0 Hz), 131.1  $(q, J_{C-F} = 3.4 \text{ Hz})$ , 126.1, 122.7  $(q, J_{C-F} = 273.4 \text{ Hz})$ , 119.5  $(q, J_{C-F} = 1.4 \text{ Hz})$ 4.0 Hz), 65.2, 55.3, 38.6, 30.3, 23.6, 19.0, 13.6; <sup>19</sup>F NMR (CDCl<sub>3</sub>, 376 MHz)  $\delta$  –62.8; IR  $\rm (cm^{-1})$   $\nu$  3672, 2971, 2876, 1719, 1329, 1263, 1188, 1084, 591; HRMS (ESI) calcd for  $C_{16}H_{19}F_3NO_5S$   $[M + H]$ <sup>+</sup> 394.0931, found 394.0933.

Ethyl (N-Acetyl-1,1-dioxo-7-methylbenzo[d]isothiazol-3-yl) acetate  $(4c)$ . This compound was obtained as a yellow oil  $(61.0)$ mg, 65%) by following general procedure A except 2.0 equiv of 2a was used: <sup>1</sup>H NMR (CDCl<sub>3</sub>, 400 MHz)  $\delta$  7.54 (t, J = 7.7 Hz, 1H), 7.34 (t,  $J = 7.2$  Hz, 2H), 5.65 (dd,  $J = 7.6$ , 3.3 Hz, 1H), 4.11 (q,  $J = 7.1$  Hz, 2H), 3.12 (dd, J = 16.0, 3.4 Hz, 1H), 2.91 (dd, J = 16.0, 7.7 Hz, 1H), 2.64 (s, 3H), 2.60 (s, 3H), 1.17 (t, J = 7.1 Hz, 3H); <sup>13</sup>C{<sup>1</sup>H} NMR  $(CDCl<sub>3</sub>, 100 MHz)$  δ 169.4, 167.3, 135.0, 134.6, 134.0, 131.9, 131.3, 121.9, 60.9, 54.7, 39.0, 23.4, 16.7, 13.8; IR (cm<sup>-1</sup>)  $\nu$  2984, 2930, 1731, 1689, 1370, 1275, 1158, 1100, 1030, 797, 598; HRMS (ESI) calcd for  $C_{14}H_{18}NO_5S$  [M + H]<sup>+</sup> 312.0900, found 312.0905.

Ethyl (N-Acetyl-1,1-dioxo-7-trifluoromethoxybenzo[d]isothiazol-3-yl)acetate (4d). This compound was obtained as a white solid (100.5 mg, 88%) by following general procedure A except 5.0 mol % of  $[Cp*RhCl<sub>2</sub>]$ <sub>2</sub> was used: mp 68–70 °C; <sup>1</sup>H NMR (CDCl<sub>3</sub>, 400 MHz)  $\delta$  7.73 (t, J = 8.0 Hz, 1H), 7.50 (d, J = 7.7 Hz, 1H), 7.43 (d, J = 8.2 Hz, 1H), 5.72 (d, J = 7.8 Hz, 1H), 4.13 (q, J = 6.9 Hz, 2H), 3.17  $(d, J = 16.2 \text{ Hz}, 1H), 2.94 \text{ (dd, } J = 16.2, 8.0 \text{ Hz}, 1H), 2.60 \text{ (s, 3H)}, 1.19$  $(t, J = 7.1 \text{ Hz}, 3\text{H})$ ; <sup>13</sup>C{<sup>1</sup>H} NMR (CDCl<sub>3</sub>, 100 MHz)  $\delta$  169.4, 167.2, 144.0, 138.0, 136.0, 125.5, 122.5, 120.1 (d,  $J_{C-F} = 262.9$ ), 119.5, 61.2, 54.9, 38.9, 23.6, 13.9; <sup>19</sup>F NMR (CDCl<sub>3</sub>, 376 MHz)  $\delta$  -57.4; IR (cm<sup>−</sup><sup>1</sup> ) 3108, 3000, 2930, 1747, 1706, 1490, 1341, 1183, 1113, 1038, 914, 797, 578; HRMS (ESI) calcd for  $C_{14}H_{15}F_3NO_6S$   $[M + H]^+$ 382.0567, found 382.0569.

Ethyl (N-Acetyl-1,1-dioxo-7-chlorobenzo[d]isothiazol-3-yl) acetate (4e). This compound was obtained as a colorless oil (96.1 mg, 97%) by following general procedure A except 2.0 equiv of 2a was used: <sup>1</sup>H NMR (CDCl<sub>3</sub>, 400 MHz)  $\delta$  7.62 (t, J = 7.9 Hz, 1H), 7.53 (d,  $J = 7.9$  Hz, 1H), 7.49 (d,  $J = 7.8$  Hz, 1H), 5.68 (dd,  $J = 7.8$ , 3.2 Hz, 1H), 4.12 (q,  $J = 7.1$  Hz, 2H), 3.14 (dd,  $J = 16.2$ , 3.3 Hz, 1H), 2.93 (dd, J = 16.2, 7.9 Hz, 1H), 2.61 (s, 3H), 1.19 (t, J = 7.1 Hz, 3H);<br><sup>13</sup>C{<sup>1</sup>H} NMR (CDCl<sub>3</sub>, 100 MHz)  $\delta$  169.2, 167.1, 137.6, 135.2, 131.4, 130.7, 128.9, 123.0, 61.0, 54.3, 38.7, 23.5, 13.8; IR (cm<sup>-1</sup>) ν 3083, 3000, 1739, 1706, 1457, 1333, 1275, 1175, 1026, 972, 802, 594; HRMS (ESI) calcd for  $C_{13}H_{15}CINO_5S$   $[M + H]^+$  332.0354, found 332.0356.

Ethyl (N-Acetyl-1,1-dioxo-7-(methoxycarbonyl)benzo[d] isothiazol-3-yl)acetate (4f). This compound was obtained as a white solid (97.7 mg, 92%) by following general procedure A except using 5.0 mol % of  $[Cp*RhCl<sub>2</sub>]$ <sub>2</sub> at 120 °C: mp 119–121 °C; <sup>1</sup>H NMR (CDCl<sub>3</sub>, 400 MHz)  $\delta$  8.22 (d, J = 7.3 Hz, 1H), 7.82 (d, J = 7.5 Hz, 1H), 7.78 (t, J = 7.6 Hz, 1H), 5.71 (d, J = 7.6 Hz, 1H), 4.12 (q, J = 7.0 Hz, 2H), 4.04 (s, 3H), 3.16 (d,  $J = 16.1$  Hz, 1H), 2.93 (dd,  $J = 16.2$ , 7.8 Hz, 1H), 2.63 (s, 3H), 1.19 (t,  $J = 7.0$  Hz, 3H); <sup>13</sup>C{<sup>1</sup>H} NMR (CDCl3, 100 MHz) δ 169.5, 167.6, 162.9, 137.2, 134.1, 133.8, 131.9, 129.5, 126.6, 61.1, 54.0, 53.1, 39.0, 23.5, 13.9; IR (cm<sup>-1</sup>)  $\nu$  3440, 2979, 2946, 2905, 1731, 1685, 1395, 1283, 1154, 1001, 773, 586; HRMS (ESI) calcd for  $C_{15}H_{18}NO_7S$  [M + H]<sup>+</sup> 356.0799, found 356.0803.

Methyl (N-Acetyl-1,1-dioxo-5-methyl-(E)-7-(methoxy-3-oxoprop-1-enyl)benzo[d]isothiazol-3-yl)acetate (30). This compound was obtained as a white solid (93.7 mg, 82%) by following general procedure A: mp 139−141 °C; <sup>1</sup>H NMR (CDCl<sub>3</sub>, 400 MHz)  $\delta$  8.02  $(d, J = 15.9 \text{ Hz}, 1\text{H}), 7.55 \text{ (s, 1H)}, 7.33 \text{ (s, 1H)}, 6.63 \text{ (d, } J = 15.9 \text{ Hz},$ 1H), 5.65 (dd, J = 7.6, 3.3 Hz, 1H), 3.83 (s, 3H), 3.69 (s, 3H), 3.13  $(dd, J = 16.2, 3.4 Hz, 1H), 2.95 (dd, J = 16.2, 7.7 Hz, 1H), 2.61 (s,$ 3H), 2.48 (s, 3H); <sup>13</sup>C{<sup>1</sup>H} NMR (CDCl<sub>3</sub>, 100 MHz)  $\delta$  170.0, 167.3, 165.8, 145.7, 136.0, 135.8, 130.7, 129.4, 128.7, 126.0, 123.9, 54.6, 52.0, 38.8, 23.5, 21.8 (one signal missing due to overlap); IR  $\rm (cm^{-1})$   $\nu$  3407, 2955, 1743, 1710, 1635, 1432, 1275, 1229, 1145.9, 968, 856, 673, 623; HRMS (ESI) calcd for  $C_{17}H_{20}NO_7S$   $[M + H]^+$  382.0955, found 382.0958.

Butyl (N-Acetyl-1,1-dioxo-5-methyl-(E)-7-(butoxy-3-oxoprop-1 enyl)benzo[d]isothiazol-3-yl)acetate (3p). This compound was obtained as a light yellow solid (130.0 mg, 93%) by following general procedure A: mp 77−78 °C; <sup>1</sup> H NMR (CDCl3, 400 MHz) δ 8.01 (d, J  $= 15.9$  Hz, 1H), 7.55 (s, 1H), 7.33 (s, 1H), 6.62 (d, J = 15.9 Hz, 1H), 5.64 (d,  $J = 5.2$  Hz, 1H), 4.22 (t,  $J = 6.6$  Hz, 2H), 4.07 (t,  $J = 6.6$  Hz,  $2H$ ), 3.13 (dd,  $I = 16.1$ , 2.6 Hz, 1H), 2.93 (dd,  $I = 16.0$ , 7.8 Hz, 1H), 2.60 (s, 3H), 2.47 (s, 3H), 1.72−1.65 (m, 2H), 1.57−1.50 (m, 2H), 1.47−1.38 (m, 2H), 1.32−1.23 (m, 2H), 0.95 (t, J = 7.3 Hz, 3H), 0.88  $(t, J = 7.3 \text{ Hz}, 3\text{H})$ ; <sup>13</sup>C{<sup>1</sup>H} NMR (CDCl<sub>3</sub>, 100 MHz)  $\delta$  169.5, 167.2, 165.3, 145.5, 136.1, 135.5, 130.7, 129.3, 128.6, 125.9, 124.2, 64.8, 64.7, 54.6, 38.8, 30.5, 30.2, 23.4, 21.7, 19.0, 18.8, 13.6, 13.5; IR (cm<sup>-1</sup>) ν 3411, 2963, 2867, 1731, 1710, 1635, 1316, 1283, 1187, 1150, 1063, 980, 864, 615; HRMS (ESI) calcd for  $C_{23}H_{32}NO_7S$   $[M + H]^+$ 466.1894, found 466.1903.

Benzyl (N-Acetyl-1,1-dioxo-5-methyl-(E)-7-(benzoxy-3-oxoprop-1-enyl)benzo[d]isothiazol-3-yl)acetate  $(3q)$ . This compound was obtained as a light yellow solid (148.7 mg, 93%) by following general procedure A: mp 134−136 °C; <sup>1</sup>H NMR (CDCl<sub>3</sub>, 400 MHz)  $\delta$  8.04  $(d, J = 15.9 \text{ Hz}, 1\text{H}), 7.46 \text{ (s, 1H)}, 7.42-7.34 \text{ (m, 5H)}, 7.31 \text{ (dd, } J =$ 4.8, 2.9 Hz, 3H), 7.29−7.25 (m, 2H), 7.19 (s, 1H), 6.63 (d, J = 15.9 Hz, 1H), 5.64 (dd, J = 7.6, 3.3 Hz, 1H), 5.26 (s, 2H), 5.10 (d, J = 1.9 Hz, 2H), 3.17 (dd, J = 15.9, 3.4 Hz, 1H), 2.98 (dd, J = 15.9, 7.7 Hz, 1H), 2.57 (s, 3H), 2.36 (s, 3H); <sup>13</sup>C{<sup>1</sup>H} NMR (CDCl<sub>3</sub>, 100 MHz)  $\delta$ 169.2, 167.4, 165.2, 145.6, 136.3, 136.0, 135.7, 135.2, 130.7, 129.5, 128.8, 128.57, 128.53, 128.51, 128.4, 128.30, 128.28, 126.0, 124.0, 66.9, 66.7, 54.7, 39.0, 23.6, 21.8 IR  $\text{(cm}^{-1})$   $\nu$  3067, 2959, 1731, 1706, 1374, 1308, 1275, 1221, 1167, 963, 748, 702; HRMS (ESI) calcd for  $C_{29}H_{31}N_2O_7S$  [M + NH<sub>4</sub>]<sup>+</sup> 551.1847, found 551.1852.

tert-Butyl (N-Acetyl-1,1-dioxo-5-methyl-(E)-7-(tert-butoxy-3-oxoprop-1-enyl)benzo[d]isothiazol-3-yl)acetate (3r). This compound was obtained as a white solid (109.0 mg, 78%) by following general procedure A except for performing the reaction at 120 °C: mp 132− 134 °C; <sup>1</sup>H NMR (CDCl<sub>3</sub>, 400 MHz)  $\delta$  7.93 (d, J = 15.8 Hz, 1H), 7.53 (s, 1H), 7.35 (s, 1H), 6.54 (d, J = 15.8 Hz, 1H), 5.62 (dd, J = 8.3, 3.1 Hz, 1H), 3.12 (dd,  $J = 15.9$ , 3.2 Hz, 1H), 2.75 (dd,  $J = 15.9$ , 8.4 Hz, 1H), 2.60 (s, 3H), 2.45 (s, 3H), 1.53 (s, 9H), 1.40 (s, 9H); <sup>13</sup>C{<sup>1</sup>H} NMR (CDCl<sub>3</sub>, 100 MHz) δ 168.9, 167.2, 164.6, 145.3, 136.3, 134.7, 131.0, 129.3, 128.4, 126.1, 125.9, 81.8, 81.2, 54.8, 40.2, 28.0, 27.9, 23.6, 21.8; IR (cm<sup>-1</sup>) *ν* 3411, 2992, 2934, 1723 1702, 1378, 1283, 11543, 972, 843, 607, 520; HRMS (ESI) calcd for  $C_{23}H_{35}N_2O_7S$   $[M + NH_4]^+$ 483.2159, found 483.2167.

N,N-Dimethyl(N-acetyl-1,1-dioxo-5-methylbenzo[d]isothiazol-3 yl) acetamide (4g). This compound was obtained as a white solid (58.6 mg, 63%) by following general procedure A: mp 145−<sup>147</sup> °C; <sup>1</sup> <sup>1</sup>H NMR (CDCl<sub>3</sub>, 400 MHz)  $\delta$  7.66 (d, J = 8.1 Hz, 1H), 7.62 (s, 1H), 7.36 (d, J = 8.0 Hz, 1H), 5.87 (d, J = 9.6 Hz, 1H), 3.24 (dd, J = 16.1,

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2.0 Hz, 1H), 2.98 (s, 3H), 2.91 (s, 3H), 2.67 (dd, J = 16.2, 9.6 Hz, 1H), 2.59 (s, 3H), 2.44 (s, 3H); <sup>13</sup>C{<sup>1</sup>H} NMR (CDCl<sub>3</sub>, 100 MHz)  $\delta$ 168.9, 167.3, 145.5, 136.4, 130.7, 130.5, 126.4, 121.1, 55.7, 38.7, 37.0, 35.5, 23.7, 22.0; IR (cm<sup>-1</sup>) ν 2926, 1706, 1649, 1395, 1312, 1263, 1147, 1022, 811, 678, 599; HRMS (ESI) calcd for  $C_{14}H_{19}N_2O_4S$  [M + H]+ 311.1060, found 311.1065.

Preparation of the Product 6 or 7 (General Procedure B). A mixture of sulfonamide 1 (0.3 mmol, 1.0 equiv),  $[Cp*RhCl<sub>2</sub>]<sub>2</sub>$  (9.27 mg, 0.015 mmol, 5.0 mol %), and AgOAc (200.4 mg, 1.2 mmol, 4.0 equiv) were weighed in a 50 mL Schlenk tube equipped with a stir bar. Dry t-AmOH (2.0 mL) was added, followed immediately by the styrene 2 (1.2 mmol, 4.0 equiv), and the mixture was stirred at 120 °C for 36 h under an Ar atmosphere. Afterward, the vial was cooled to room temperature, diluted with  $CH_2Cl_2$ , and transferred to a roundbottom flask. Silica was added to the flask, and the volatiles were evaporated under reduced pressure. The purification was performed by flash column chromatography on silica gel (eluent EtOAc/petroleum ether 1/3, typically).

 $(E)-N-(4-Methyl-2,6-distyrylphenylsulfonyl)acetamide (6a). This$ compound was obtained as a white solid (100.8 mg, 81%) by following general procedure B: mp 132−134 °C; <sup>1</sup>H NMR (CDCl<sub>3</sub>, 400 MHz)  $\delta$ 8.62 (s, 1H), 7.87 (d, J = 16.0 Hz, 2H), 7.44 (d, J = 7.3 Hz, 4H), 7.27 (s, 2H), 7.25−7.13 (m, 6H), 6.75 (d, J = 16.0 Hz, 2H), 2.36 (s, 3H), 1.74 (s, 3H); <sup>13</sup>C{<sup>1</sup>H} NMR (DMSO- $d_6$ , 100 MHz)  $\delta$  169.4, 143.4, 139.2, 137.1, 131.3, 130.8, 129.0, 128.7, 128.0, 127.6, 127.0, 23.1, 20.8; IR (cm<sup>−</sup><sup>1</sup> ) ν 3415, 3253, 3038, 1723, 1557, 1411, 1324, 1146, 963, 847, 652, 520; HRMS (ESI) calcd for  $C_{25}H_{24}NO_3S$   $[M + H]^+$ 418.1471, found 418.1466.

(E)-N-(4-Methyl-2,6-bis(4-methylstyryl)phenylsulfonyl)acetamide **(6b).** This compound was obtained as a white solid  $(97.4 \text{ mg}, 73%)$  by following general procedure B: mp 227−229 °C; <sup>1</sup> H NMR (DMSO $d_6$ , 400 MHz)  $\delta$  12.46 (s, 1H), 7.91 (d, J = 16.0 Hz, 2H), 7.53–7.50  $(m, 6H)$ , 7.22 (d, J = 7.7 Hz, 4H), 6.99 (d, J = 16.0 Hz, 2H), 2.42 (s, 3H), 2.33 (s, 6H), 1.92 (s, 3H); <sup>13</sup>C{<sup>1</sup>H} NMR (DMSO- $d_6$ , 100 MHz) δ 169.4, 143.3, 139.3, 137.5, 134.4, 131.2, 130.6, 129.3, 128.8, 126.9, 126.6, 23.2, 20.95, 20.88; IR (cm<sup>-1</sup>)  $\nu$  3689, 3166, 2884, 2689, 1664, 1449, 1341, 1158, 847, 648; HRMS (ESI) calcd for  $C_{27}H_{28}NO_3S$  $[M + H]$ <sup>+</sup> 446.1784, found 446.1779.

(E)-N-(2,6-Bis(4-tert-butylstyryl)-4-methylphenylsulfonyl) acetamide (6c). This compound was obtained as a white solid (126.6 mg, 80%) by following general procedure B: mp 207–209 °C; <sup>1</sup>H NMR (DMSO- $d_6$ , 400 MHz)  $\delta$  12.43 (s, 1H), 7.93 (d, J = 16.1 Hz, 2H), 7.54 (d,  $J = 7.6$  Hz, 6H), 7.43 (d,  $J = 8.2$  Hz, 4H), 7.01 (d,  $J =$ 16.0 Hz, 2H), 2.42 (s, 3H), 1.92 (s, 3H), 1.30 (s, 18H);  $^{13}C(^{1}H)$ NMR (DMSO- $d_6$ , 100 MHz)  $\delta$  169.4, 150.6, 143.3, 139.3, 134.4, 131.1, 130.6, 128.7, 126.89, 126.76, 125.5, 34.4, 31.1, 23.2, 20.9; IR (cm<sup>−</sup><sup>1</sup> ) ν 3158, 2959, 2863, 1685, 1457, 1349, 1163, 9768, 851, 661, 532; HRMS (ESI) calcd for  $C_{33}H_{43}N_2O_3S$   $[M + NH_4]^+$  547.2989, found 547.2995.

(E)-N-(2,6-Bis(4-methoxystyryl)-4-methylphenylsulfonyl) acetamide (6d). This compound was obtained as a yellow solid (109.2) mg, 76%) by following general procedure B: mp 224−226 °C; <sup>1</sup>H NMR (DMSO- $d_6$ , 400 MHz)  $\delta$  12.41 (s, 1H), 7.81 (d, J = 16.1 Hz, 2H), 7.55 (d, J = 8.1 Hz, 4H), 7.50 (s, 2H), 7.02−6.95 (m, 6H), 3.79  $(s, 6H)$ , 2.41  $(s, 3H)$ , 1.91  $(s, 3H)$ ; <sup>13</sup>C{<sup>1</sup>H} NMR (DMSO- $d_6$ , 100 MHz) δ 169.3, 159.2, 143.1, 139.4, 130.8, 130.3, 129.8, 128.4, 128.3, 125.3, 114.1, 55.2, 23.1, 20.8; IR (cm<sup>-1</sup>)  $\nu$  3262, 2934, 2834, 1723, 1615, 1507, 1440, 1245, 1133, 1026, 847, 644, 532; HRMS (ESI) calcd for  $C_{27}H_{31}N_2O_5S$   $[M + NH_4]^+$  495.1948, found 495.1947.

(E)-N-(2,6-Bis(4-fluorostyryl)-4-methylphenylsulfonyl)acetamide (6e). This compound was obtained as a yellow solid (118.2 mg, 87%) by following general procedure B: mp 257−259 °C; <sup>1</sup> H NMR  $(DMSO-d<sub>6</sub>, 400 MHz) \delta 12.49$  (s, 1H), 7.90 (d, J = 16.0 Hz, 2H), 7.66 (m, 4H), 7.54 (s, 2H), 7.26 (t, J = 7.9 Hz, 4H), 7.04 (d, J = 16.0 Hz, 2H), 2.42 (s, 3H), 1.93 (s, 3H); <sup>13</sup>C{<sup>1</sup>H} NMR (DMSO- $d_6$ , 100 MHz)  $\delta$  169.5, 161.9 (d, J<sub>C−F</sub> = 244.9 Hz), 143.4, 139.1, 133.7 (d, J<sub>C−F</sub>  $= 3.0$  Hz), 130.8, 130.1, 129.0, 128.9 (d,  $J_{C-F} = 8.1$  Hz), 127.5, 115.6 (d,  $J_{C-F}$  = 21.6 Hz), 23.2, 20.9; <sup>19</sup>F NMR (DMSO- $d_6$ , 376 MHz)  $\delta$ −113.8; IR (cm<sup>−</sup><sup>1</sup> ) ν 3245, 2926, 1731, 1507, 1325, 1221, 11475, 860,

657; HRMS (ESI) calcd for  $C_{25}H_{25}F_2N_2O_3S$   $[M + NH_4]^+$  471.1549, found 471.1553.

(E)-N-(2,6-Bis(4-chlorostyryl)-4-methylphenylsulfonyl)acetamide (6f). This compound was obtained as a white solid  $(135.6 \text{ mg}, 93\%)$  by following general procedure B: mp 253−255 °C; <sup>1</sup>H NMR (CDCl<sub>3</sub>, 400 MHz)  $\delta$  8.73 (s, 1H), 7.93 (d, J = 16.0 Hz, 2H), 7.45 (d, J = 8.5 Hz, 4H), 7.36 (s, 2H), 7.29 (d, J = 8.4 Hz, 4H), 6.79 (d, J = 16.0 Hz, 2H), 2.46 (s, 3H), 1.88 (s, 3H); <sup>13</sup>C{<sup>1</sup>H} NMR (DMSO- $d_6$ , 100 MHz) δ 169.5, 143.5, 138.9, 136.0, 132.4, 130.8, 130.0, 129.2, 128.7, 128.6, 128.3, 23.2, 20.8; IR (cm<sup>-1</sup>)  $\nu$  3419, 3100, 2901, 1669, 1482, 1333, 1158, 1075, 1009, 831, 656, 520; HRMS (ESI) calcd for  $C_{25}H_{22}Cl_2NO_3S$  [M + H]<sup>+</sup> 486.0692, found 486.0702.

(E)-N-(2,6-Bis(4-bromostyryl)-4-methylphenylsulfonyl)acetamide (6g). This compound was obtained as a white solid (162.5 mg, 94%) by following general procedure B: mp 231−233 °C; <sup>1</sup> H NMR  $(DMSO-d<sub>6</sub>, 400 MHz)$  δ 12.52 (s, 1H), 7.97 (d, J = 16.1 Hz, 2H), 7.62  $(d, J = 8.5$  Hz, 4H), 7.58–7.54 (m, 6H), 7.02 (d,  $J = 16.1$  Hz, 2H), 2.42 (s, 3H), 1.93 (s, 3H); <sup>13</sup>C{<sup>1</sup>H} NMR (DMSO- $d_6$ , 100 MHz)  $\delta$ 169.5, 143.5, 138.9, 136.4, 131.7, 130.8, 130.1, 129.2, 128.9, 128.4, 121.1, 23.2, 20.9; IR (cm<sup>-1</sup>)  $\nu$  3436, 3374, 1706, 1491, 1147, 997, 653, 520; HRMS (ESI) calcd for  $C_{25}H_{25}Br_2N_2O_3S$   $[M + NH_4]^+$  592.9927, found 592.9923.

(E)-N-(4-Methyl-2,6-bis(perfluorostyryl)phenylsulfonyl)acetamide (6h). This compound was obtained as a brown solid (106.2 mg, 59%) by following general procedure B: mp 196−198 °C; <sup>1</sup> H NMR  $(DMSO-d<sub>6</sub>, 400 MHz)$  δ 12.36 (s, 1H), 8.23 (d, J = 16.4 Hz, 2H), 7.63  $(s, 2H)$ , 6.83 (d, J = 16.5 Hz, 2H), 2.46 (s, 3H), 1.85 (s, 3H); <sup>13</sup>C{<sup>1</sup>H} NMR (DMSO-d<sub>6</sub>, 100 MHz) δ 170.1, 145.9–145.7 (m), 144.6, 143.5−143.2 (m), 138.9−138.6 (m), 138.4, 136.8−136.2 (m), 130.5, 128.8 (d,  $J_{C-F}$  = 205.2 Hz), 116.0, 112.5−112.2 (m), 23.2, 21.0; <sup>19</sup>F NMR (DMSO- $d_6$ , 376 MHz)  $\delta$  -137.74 (dd, J = 23.3, 7.2 Hz),  $\delta$  $-151.10$  (t, J = 22.1 Hz),  $-158.34$  (td, J = 23.0, 7.3 Hz); IR (cm<sup>-1</sup>)  $\nu$ 3647, 3237, 2930, 1731, 1523, 1499, 1337, 1146, 1005, 847, 656; HRMS (ESI) calcd for  $C_{25}H_{14}F_{10}NO_3S$  [M + H]<sup>+</sup> 598.0529, found 598.0525.

 $(E)-N-(4-Chloro-2,6-distryly phenylsulfonyl)acetamide (6i). This$ compound was obtained as a white solid (88.1 mg, 67%) by following general procedure B: mp 227−229 °C; <sup>1</sup>H NMR (DMSO- $d_6$ , 400 MHz)  $\delta$  12.67 (s, 1H), 7.93 (d, J = 16.1 Hz, 2H), 7.81 (s, 2H), 7.64 (d,  $J = 7.4$  Hz, 4H), 7.43 (t,  $J = 7.6$  Hz, 4H), 7.34 (t,  $J = 7.3$  Hz, 2H), 7.17  $(d, J = 16.1 \text{ Hz}, 2\text{H})$ , 1.96 (s, 3H); <sup>13</sup>C{<sup>1</sup>H} NMR (DMSO- $d_6$ , 100 MHz) δ 169.7, 141.4, 138.0, 136.8, 132.9, 132.1, 128.7, 128.4, 127.4, 127.2, 126.1, 23.1; IR (cm<sup>-1</sup>)  $\nu$  3324, 3017, 1715, 1545, 1350, 1155, 840, 686, 524; HRMS (ESI) calcd for  $C_{24}H_{24}CIN_2O_3S$   $[M + NH_4]^+$ 455.1191, found 455.1197.

(E)-N-(4-Methoxy-2,6-distyrylphenylsulfonyl)acetamide (6j). This compound was obtained as a red solid (72.2 mg, 56%) by following general procedure B: mp 204−206 °C; <sup>1</sup>H NMR (CDCl<sub>3</sub>, 400 MHz)  $\delta$ 8.99 (s, 1H), 7.95 (d, J = 16.0 Hz, 2H), 7.52 (d, J = 7.5 Hz, 4H), 7.32  $(t, J = 7.5 \text{ Hz}, 4\text{H})$ , 7.23  $(t, J = 3.6 \text{ Hz}, 2\text{H})$ , 7.02  $(s, 2\text{H})$ , 6.83  $(d, J =$ 16.0 Hz, 2H), 3.91 (s, 3H), 1.80 (s, 3H); <sup>13</sup>C{<sup>1</sup>H} NMR (CDCl<sub>3</sub>, 100 MHz) δ 168.9, 162.5, 142.7, 136.6, 133.0, 128.7, 128.2, 127.6, 127.0, 114.0, 55.7, 23.0; IR (cm<sup>-1</sup>) ν 3245, 3021, 1723, 1586, 1411, 1337, 1146, 959, 847, 681, 524; HRMS (ESI) calcd for  $C_{25}H_{27}N_2O_4S$  [M + NH<sub>4</sub><sup>+</sup> 451.1686, found 451.1682.

(E)-N-(3-Bromo-2,6-distyrylphenylsulfonyl)acetamide (6k). This compound was obtained as a light yellow solid (43.1 mg, 30%) by following general procedure B: mp 165−167 °C; <sup>1</sup>H NMR (CDCl<sub>3</sub>, 400 MHz)  $\delta$  8.07 (s, 1H), 8.00 (d, J = 16.1 Hz, 1H), 7.86 (d, J = 8.4 Hz, 1H), 7.55 (d, J = 7.2 Hz, 4H), 7.47 (d, J = 8.4 Hz, 1H), 7.43–7.27  $(m, 7H)$ , 6.88 (d, J = 16.1 Hz, 1H), 6.59 (d, J = 16.7 Hz, 1H), 1.90 (s, 3H); <sup>13</sup>C{<sup>1</sup>H} NMR (CDCl<sub>3</sub>, 100 MHz)  $\delta$  168.0, 139.9, 139.7, 137.5, 136.7, 136.4, 136.0, 133.1, 129.7, 129.0, 128.8, 128.7, 128.44, 128.40, 127.1, 126.9, 126.8, 126.2, 125.0, 23.3; IR (cm<sup>-1</sup>)  $\nu$  3058, 2360, 1686, 1558, 1496, 1457, 1348, 1166, 859, 756, 605; HRMS (ESI) calcd for  $C_{24}H_{21}BrNO_3S$  [M + H]<sup>+</sup> 482.0420, found 482.0414.

(E)-N-(2-Styryl-5-(trifluoromethyl)phenylsulfonyl)acetamide (7a). This compound was obtained as brown solid (95.8 mg, 86%) by following general procedure B: mp 165−167 °C; <sup>1</sup>H NMR (CDCl<sub>3</sub>, 400 MHz)  $\delta$  9.26 (s, 1H), 8.42 (s, 1H), 7.82 (d, J = 13.2 Hz, 3H), 7.52

<span id="page-8-0"></span> $(d, J = 5.1 \text{ Hz}, 2H), 7.30 \text{ (s, 3H)}, 7.09 \text{ (d, } J = 15.9 \text{ Hz}, 1H), 1.95 \text{ (s, }$ 3H); <sup>13</sup>C{<sup>1</sup>H} NMR (CDCl<sub>3</sub>, 100 MHz)  $\delta$  168.8, 140.7, 136.5, 135.8, 135.6, 130.6 (d,  $J_{C-F}$  = 3.2 Hz), 129.5 (d,  $J_{C-F}$  =34.1 Hz), 129.2, 128.8, 128.7, 128.2 (d,  $J_{C-F}$  = 3.5 Hz), 127.4, 123.1 (d,  $J_{C-F}$  = 272.6 Hz), 122.5, 23.1; <sup>19</sup>F NMR (CDCl<sub>3</sub>, 376 MHz)  $\delta$  –62.6; IR (cm<sup>-1</sup>)  $\nu$  3087, 2884, 1681, 1615, 1474, 1328, 1158, 1129, 1080, 885, 615; HRMS (ESI) calcd for  $C_{17}H_{15}F_3NO_3S$  [M + H]<sup>+</sup> 370.0719, found 370.0717.

Synthetic Transformations of 3a and 6a. Methyl (1,1-Dioxo-2,3-dihydro-5-methyl-(E)-7-(methoxy-3-oxoprop-1-enyl)benzo[d] isothiazol-3-yl)acetate  $(8)$  (Eq 1). A mixture of 3a  $(61.4 \text{ mg}, 0.15$ mmol),  $K_2CO_3$  (22.8 mg, 0.165 mmol, 1.1 equiv) and MeOH/H<sub>2</sub>O 20/1 (3.0 mL) was stirred for 0.5 h at room temperature. Methanol was evaporated, and the residue [w](#page-3-0)as diluted with EtOAc (25.0 mL) and then washed with brine (20.0 mL). The organic layer was dried over  $\text{Na}_2\text{SO}_4$  and concentrated in vacuo. The residue was purified by column chromatography (eluent EtOAc/petroleum ether 1/2) to afford 40.0 mg of 8 in 78% yield as a white solid: mp 123−125 °C; <sup>1</sup>H NMR (CDCl<sub>3</sub>, 400 MHz)  $\delta$  8.04 (d, J = 15.8 Hz, 1H), 7.50 (s, 1H), 7.15 (s, 1H), 6.62 (d, J = 15.9 Hz, 1H), 5.58 (d, J = 21.5 Hz, 1H), 5.03  $(br s, 1H)$ , 3.81  $(s, 3H)$ , 3.76  $(s, 3H)$ , 2.95  $(d, J = 17.0 \text{ Hz}, 1H)$ , 2.79 (dd, J = 16.8, 10.0 Hz, 1H), 2.46 (s, 3H); <sup>13</sup>C{<sup>1</sup>H} NMR (CDCl<sub>3</sub>, 100 MHz) δ 171.1, 166.1, 144.6, 139.6, 136.6, 131.7, 130.4, 128.4, 125.4, 123.4, 52.9, 52.4 52.0, 40.2, 21.7; IR (cm<sup>-1</sup>)  $\nu$  3217, 2959, 1712, 1443, 1290, 1179, 1147, 970, 859, 669; HRMS (ESI) calcd for  $C_1,H_2,N_2O_6S$  $[M + NH<sub>4</sub>]$ <sup>+</sup> 357.1114, found 357.1120.

Methyl (1,1-Dioxo-2,3-dihydro-5-methyl-7-(methoxy-3 oxopropanyl)benzo[d]Isothiazol-3-yl)acetate (9) (Eq 2). To a solution of  $8$  (96 mg, 0.3 mmol) in THF  $(1.0 \text{ mL})$  and MeOH  $(0.7 \text{ m})$ mL) were added  $NiCl<sub>2</sub>·H<sub>2</sub>O$  (142.6 mg, 0.6 mmol, 2.0 equiv) and NaBH<sub>4</sub> (56.7 mg, 1.5 mmol, 5.0 equiv) at 0  $\degree$ C und[er](#page-3-0) an argon atmosphere. The resulting mixture was stirred overnight and then quenched with water (3.0 mL). The crude mixture was diluted with EtOAc and filtered through a pad of Celite. The filtrate was separated, and the organic layer was washed with water and brine and dried over  $Na<sub>2</sub>SO<sub>4</sub>$ . The solvent was evaporated, and the crude mixture was purified by column chromatography (eluent EtOAc/petroleum ether 1/3) to afford 68.8 mg of 9 in 73% yield as a white solid: mp 91−93  $^{\circ}$ C; <sup>1</sup>H NMR (CDCl<sub>3</sub>, 400 MHz)  $\delta$  7.14 (s, 1H), 6.98 (s, 1H), 5.69 (d, J = 17.2 Hz, 1H), 5.03−4.96 (m, 1H), 3.72 (s, 3H), 3.64 (s, 3H), 3.22 (t, J = 7.6 Hz, 2H), 2.92 (dd, J = 16.9, 3.0 Hz, 1H), 2.78−2.70 (m, 3H), 2.38 (s, 3H); <sup>13</sup>C{<sup>1</sup>H} NMR (CDCl<sub>3</sub>, 100 MHz)  $\delta$  172.7, 171.2, 144.4, 138.9, 136.7, 131.4, 131.3, 122.3, 52.9, 52.2, 51.6, 40.3, 34.3, 26.2, 21.6; IR (cm<sup>−1</sup>) ν 3249, 2967, 1718, 1449, 1291, 1146, 992, 851, 665, 532; HRMS (ESI) calcd for  $C_{15}H_{20}NO_6S$   $[M + H]^+$  342.1006, found 342.1003.

N-(4-Methyl-2,6-diphenethylphenylsulfonyl)acetamide (10) (Eq 2). To a solution of 6a (125.3 mg, 0.3 mmol) in THF (1.0 mL) and MeOH  $(0.7 \text{ mL})$  were added NiCl<sub>2</sub>·H<sub>2</sub>O  $(142.6 \text{ mg}, 0.6 \text{ mmol}, 2.0 \text{ m}$ equiv) and NaBH<sub>4</sub> (56.7 mg, 1.5 mmol, 5.0 equiv) at 0  $\degree$ C under an [ar](#page-3-0)gon atmosphere. The resulting mixture was stirred overnight and then quenched with water (3.0 mL). The crude mixture was diluted with EtOAc and filtered through a pad of Celite. The filtrate was separated, and the organic layer was washed with water and brine and dried over  $Na<sub>2</sub>SO<sub>4</sub>$ . The solvent was evaporated, and the crude mixture was purified by column chromatography (eluent EtOAc/petroleum ether 1/2) to afford 94.2 mg of 10 in 75% yield as a white solid: mp 156−158 °C; <sup>1</sup> H NMR (CDCl3, 400 MHz) δ 8.99 (s, 1H), 7.35−7.29  $(m, 8H)$ , 7.22 (dd, J = 10.7, 6.4 Hz, 2H), 7.02 (s, 2H), 3.36 (dd, J = 9.8, 6.7 Hz, 4H), 3.01 (dd, J = 9.8, 6.6 Hz, 4H), 2.33 (s, 3H), 1.93 (s, 3H); <sup>13</sup>C{<sup>1</sup>H} NMR (CDCl<sub>3</sub>, 100 MHz)  $\delta$  168.8, 144.2, 143.8, 141.7, 131.9, 128.6, 128.3, 126.0, 38.6, 37.4, 23.3, 21.1; IR (cm<sup>-1</sup>) ν 3104, 3025, 2909, 2850, 1686, 1445, 1349, 1167, 831, 656, 511; HRMS (ESI) calcd for  $C_{25}H_{28}NO_3S$  [M + H]<sup>+</sup> 422.1784, found 422.1787.

Deuteration Experiments. Scheme 3a. The reaction between 1a (0.15 mmol) and  $D_2O$  (15 equiv) was carried out in the presence of  $[Cp*RhCl<sub>2</sub>]<sub>2</sub>$  (3.0 mol %) and Cu(OAc)<sub>2</sub>·H<sub>2</sub>O (2.0 equiv) with Tol- $d_8$ (1.0 mL) as the solvent at 100 °C for 24 [h.](#page-3-0) After standard workup and purification, a mixture of 1a and D-1a was obtained. <sup>1</sup>H NMR analysis of the mixture showed 73% D was incorporated into the two ortho

positions of the sulfonamide aryl ring (see the attached spectrum in the Supporting Information).

Scheme 3b. The reaction of 1a (0.15 mmol), alkene 2a (3.0 equiv), and  $D_2O$  (15.0 equiv) was carried out in the presence of  $[Cp*RhCl_2]_2$  $(3.0 \text{ mol } \%)$  and  $Cu(OAc)<sub>2</sub>·H<sub>2</sub>O$   $(2.0 \text{ equiv})$  with Tol- $d_8$   $(1.0 \text{ mL})$  as the solvent [a](#page-3-0)t 100 °C for 2 h. Afterward, the vial was cooled to room temperature, diluted with  $CH_2Cl_2$ , and transferred to a round-bottom flask. Silica was added to the flask, and volatiles were evaporated under reduced pressure. The purification was performed by flash column chromatography (eluent EtOAc/petroleum ether 1/2) on silica gel. Compound 3a was obtained in 10% yield. The starting material 1a was recovered in 64% yield, while monoalkenylated product was obtained in 25% yield. These two compounds cannot be separated by the flash column chromatography, and the yield was determined by <sup>1</sup>H NMR. In all of these compounds, no deuterium incorporation was detected (see the attached spectrum in the Supporting Information).

## ■ ASSOCIATED CONTENT

## **S** Supporting Information

Full spectroscopic data for all new compounds, and details of the optimization. This material is available free of charge via the Internet at http://pubs.acs.org.

## ■ AUTH[OR INFORMATIO](http://pubs.acs.org)N

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#### Notes

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## ■ REFERENCES

(1) For recent selected reviews, see: (a) Arockiam, P. B.; Bruneau, C.; Dixneuf, P. H. Chem. Rev. 2012, 112, 5879. (b) Kuhl, N.; Hopkinson, M. N.; Wencel-Delord, J.; Glorius, F. Angew. Chem., Int. Ed. 2012, 51, 10236. (c) Li, B.-J.; Shi, Z.-J. Chem. Soc. Rev. 2012, 41, 5588. (d) Cho, S. H.; Kim, J. Y.; Kwak, J.; Chang, S. Chem. Soc. Rev. 2011, 40, 5068. (e) Yeung, C. S.; Dong, V. M. Chem. Rev. 2011, 111, 1215. (f) Wencel-Delord, J.; Dröge, T.; Liu, F.; Glorius, F. Chem. Soc. Rev. 2011, 40, 4740. (g) Mkhalid, I. A. I.; Barnard, J. H.; Marder, T. B.; Murphy, J. M.; Hartwig, J. F. Chem. Rev. 2010, 110, 890.

(2) For selected reviews, see: (a) Chen, X.; Engle, K. M.; Wang, D.- H.; Yu, J.-Q. Angew. Chem., Int. Ed. 2009, 48, 5094. (b) Yamaguchi, J.; Yamaguchi, A. D.; Itami, K. Angew. Chem., Int. Ed. 2012, 51, 8960. (c) Wencel-Delord, J.; Glorius, F. Nat. Chem. 2013, 5, 369.

(3) For the first report, see: (a) Moritani, I.; Fujiwara, Y. Tetrahedron Lett. 1967, 1119. For reviews, see: (b) Jia, C.; Kitamura, T.; Fujiwara, Y. Acc. Chem. Res. 2001, 34, 633. (c) Beccalli, E. M.; Broggini, G.; Martinelli, M.; Sottocornola, S. Chem. Rev. 2007, 107, 5318.

(4) For recent reviews, see: (a) The Mizoroki-Heck Reaction; Oestreich, M., Ed.; Wiley: Chichester, U.K., 2009. (b) Bräse, S.; de Meijere, A. In Metal-Catalyzed Cross-Coupling Reactions; de Meijere, A., Diederich, F., Eds.; Wiley-VCH: New York, 2004; Chapter 5.

(5) For recent examples, see: (a) Nishikata, T.; Lipshutz, B. H. Org. Lett. 2010, 12, 1972. (b) Shi, B.-F.; Zhang, Y.-H.; Lam, J. K.; Wang, D.- H.; Yu, J.-Q. J. Am. Chem. Soc. 2010, 132, 460. (c) Wasa, M.; Engle, K. M.; Yu, J.-Q. J. Am. Chem. Soc. 2010, 132, 3680. (d) Wang, D.-H.; Engle, K. M.; Shi, B.-F.; Yu, J.-Q. Science 2010, 327, 315. (e) García-Rubia, A.; Fernández-Ibáeñz, M. A.; Gómez Arrayás, R.; Carretero, J. C. Chem. Eur. J. 2011, 17, 3567. (f) Li, D.-D.; Yuan, T.-T.; Wang, G.- W. Chem. Commun. 2011, 47, 12789. (g) Gandeepan, P.; Cheng, C.-H. J. Am. Chem. Soc. 2012, 134, 5738. (h) Leow, D.; Li, G.; Mei, T.-S.; Yu,

<span id="page-9-0"></span>J.-Q. Nature 2012, 486, 518. (i) Urones, B.; Arrayás, R. G.; Carretero, J. C. Org. Lett. 2013, 15, 1120.

(6) For reviews, see: (a) Satoh, T.; Miura, M. Chem. Eur. J. 2010, 16, 11212. (b) Song, G.; Wang, F.; Li, X. Chem. Soc. Rev. 2012, 41, 3651. (c) Patureau, F. W.; Wencel-Delord, J.; Glorius, F. Aldrichim. Acta 2012, 45, 31. (d) Han, Y.-F.; Jin, G.-X. Chem. Soc. Rev. 2014, 43, 2799. (e) Kuhl, N.; Schröder, N.; Glorius, F. Adv. Synth. Catal. 2014, 356, 1443. For selected examples, see: (f) Ueura, K.; Satoh, T.; Miura, M. Org. Lett. 2007, 9, 1407. (g) Umeda, N.; Hirano, K.; Satoh, T.; Miura, M. J. Org. Chem. 2009, 74, 7094. (h) Patureau, F. W.; Glorius, F. J. Am. Chem. Soc. 2010, 132, 9982. (i) Gong, T.-J.; Xiao, B.; Liu, Z.-J.; Wan, J.; Xu, J.; Luo, D.-F.; Fu, Y.; Liu, L. Org. Lett. 2011, 13, 3235. (j) Park, S.; Kim, J. Y.; Chang, S. Org. Lett. 2011, 13, 2372. (k) Tsai, A. S.; Brasse, M.; Bergman, R. G.; Ellman, J. A. Org. Lett. 2011, 13, 540. (l) Rakshit, S.; Grohmann, C.; Besset, T.; Glorius, F. J. Am. Chem. Soc. 2011, 133, 2350. (m) Zhu, C.; Flack, J. R. Chem. Commun. 2012, 48, 1674. (n) Dong, Y.; Liu, G. Chem. Commum. 2013, 49, 8066. (o) Ryu, T.; Kim, J.; Park, Y.; Kim, S.; Lee, P. H. Org. Lett. 2013, 15, 3986. (p) Feng, C.; Feng, D.; Loh, T.-P. Org. Lett. 2013, 15, 3670. (q) Zhou, J.; Li, B.; Hu, F.; Shi, B.-F. Org. Lett. 2013, 15, 3460. (r) Unoh, Y.; Hashimoto, Y.; Takeda, D.; Hirano, K.; Satoh, T.; Miura, M. Org. Lett. 2013, 15, 3258. (s) Brasse, M.; Campora, J.; Ellman, J. A.; Bergman, R. ́ G. J. Am. Chem. Soc. 2013, 135, 6427. (t) Nobushige, K.; Hirano, K.; Satoh, T.; Miura, M. Org. Lett. 2014, 16, 1188. (u) Boultadakis-Arapinis, M.; Hopkinson, M. N.; Glorius, F. Org. Lett. 2014, 16, 1630. (v) Parthasarathy, K.; Bolm, C. Chem. Eur. J. 2014, 20, 4896.

(7) (a) Weissman, H.; Song, X.; Milstein, D. J. Am. Chem. Soc. 2001, 123, 337. (b) Ueyama, T.; Mochida, S.; Fukutani, T.; Hirano, K.; Satoh, T.; Miura, M. Org. Lett. 2011, 13, 706. (c) Kwon, K.-H.; Lee, D. W.; Yi, C. S. Organometallics 2010, 29, 5748.

(8) For reviews, see: (a) Kozhushkov, S. I.; Ackermann, L. Chem. Sci. 2013, 4, 886. For selected examples, see: (b) Ueyama, T.; Mochida, S.; Fukutani, T.; Hirano, K.; Satoh, T.; Miura, M. Org. Lett. 2011, 13, 706. (c) Ackermann, L.; Pospech, J. Org. Lett. 2011, 13, 4153. (d) Arockiam, P. B.; Fischmeister, C.; Bruneau, C.; Dixneuf, P. H. Green Chem. 2011, 13, 3075. (e) Padala, K.; Jeganmohan, M. Org. Lett. 2011, 13, 6144. (f) Flegeau, E. F.; Bruneau, C.; Dixneuf, P. H.; Jutand, A. J. Am. Chem. Soc. 2011, 133, 10161. (g) Ackermann, L.; Wang, L.; Wolfram, R.; Lygin, A. V. Org. Lett. 2012, 14, 728. (h) Li, B.; Devaraj, K.; Darcel, C.; Dixneuf, P. Green Chem. 2012, 14, 2706. (i) Padala, K.; Pimparkar, S.; Madasamy, P.; Jeganmohan, M. Chem. Commun. 2012, 48, 7140. (j) Goriya, Y.; Ramana, C. V. Chem. Eur. J. 2012, 18, 13288. (k) Chidipudi, S. R.; Wieczysty, M. D.; Khan, I.; Lam, H. W. Org. Lett. 2013, 15, 570. (l) Lanke, V.; Prabhu, K. Org. Lett. 2013, 15, 2818. (m) Li, J.; John, M.; Ackermann, L. Chem. Eur. J. 2014, 20, 5403. (n) Hashimoto, Y.; Ueyama, T.; Fukutani, T.; Hirano, K.; Satoh, T.; Miura, M. Chem. Lett. 2011, 40, 1165. (o) Hashimoto, Y.; Ortloff, T.; Hirano, K.; Satoh, T.; Bolm, C.; Miura, M. Chem. Lett. 2012, 41, 151. (9) For selected reviews, see: (a) Alberico, D.; Scott, M. E.; Lautens, M. Chem. Rev. 2007, 107, 174. (b) Lyons, T. W.; Sanford, M. S. Chem.

Rev. 2010, 110, 1147. (c) Rouquet, G.; Chatani, N. Angew. Chem., Int. Ed. 2013, 52, 11726.

(10) Kalgutkar, A.; Jones, R.; Sawant, A. In Metabolism Pharmacokinetics and Toxicity of Functional Groups; Smith, D. A., Ed.; RSC: Cambridge, U.K., 2010; RSC Drug Discovery Series No. 1, Chapter 5.

(11) (a) Drews, J. Science. 2000, 287, 1960. (b) Levy, L. Drugs Future 1992, 17, 451.

(12) Francotte, P.; De Tullio, P.; Goffin, E.; Dintilhac, G.; Graindorge, E.; Fraikin, P.; Lestage, P.; Danober, L.; Thomas, J.-Y.; Caignard, D.-H.; Pirotte, B. J. Med. Chem. 2007, 50, 3153.

(13) (a) Zhuang, L.; Wai, J. S.; Embrey, M. W.; Fisher, T. E.; Egbertson, M. S.; Payne, L. S.; Guare, J. P., Jr.; Vacca, J. P.; Hazuda, D. J.; Felock, P. J.; Wolfe, A. L.; Stillmock, K. A.; Witmer, M. V.; Moyer, G.; Schleif, W. A.; Gabryelski, L. J.; Leonard, Y. M.; Lynch, J. J., Jr.; Michelson, S. R.; Young, S. D. J. Med. Chem. 2003, 46, 453. (b) Cavasotto, C. N.; Orry, A. J. W.; Murgolo, N. J.; Czarniecki, M. F.; Kocsi, S. A.; Hawes, B. E.; O'Neil, K. A.; Hine, H.; Burton, M. S.;

Voigt, J. H.; Abagyan, R. A.; Bayne, M. L.; Monsma, F. J., Jr. J. Med. Chem. 2008, 51, 581.

(14) (a) Pham, M. V.; Ye, B.; Cramer, N. Angew. Chem., Int. Ed. 2012, 51, 10610. (b) Wang, D.; Wang, F.; Song, G.; Li, X. Angew. Chem., Int. Ed. 2012, 51, 12348.

(15) (a) Miura, M.; Tsuda, T.; Satoh, T.; Pivsa-Art, S.; Nomura, M. J. Org. Chem. 1998, 63, 5211. (b) Li, X.; Gong, X.; Zhao, M.; Song, G.; Deng, J.; Li, X. Org. Lett. 2011, 13, 5808. (c) Hu, S.; Wang, D.; Liu, J.; Li, X. Org. Biomol. Chem. 2013, 11, 2761. (d) Dai, H.-X.; Stepan, A. F.; Plummer, M. S.; Yu, J.-Q. J. Am. Chem. Soc. 2011, 133, 7222.

(16) (a) Li, B.; Wang, N.; Liang, Y.; Xu, S.; Wang, B. Org. Lett. 2013, 15, 136. (b) Li, B.; Ma, J.; Xie, W.; Song, H.; Xu, S.; Wang, B. J. Org. Chem. 2013, 78, 9345. (c) Li, B.; Ma, J.; Xie, W.; Song, H.; Xu, S.; Wang, B. Chem. Eur. J. 2013, 19, 11863. (d) Li, B.; Ma, J.; Liang, Y.; Wang, N.; Xu, S.; Song, H.; Wang, B. Eur. J. Org. Chem. 2013, 1950. (e) Wang, N.; Li, B.; Song, H.; Xu, S.; Wang, B. Chem. Eur. J. 2013, 19, 358. (f) Li, B.; Feng, H.; Wang, N.; Ma, J.; Song, H.; Xu, S.; Wang, B. Chem. Eur. J. 2012, 18, 12873. (g) Li, B.; Ma, J.; Wang, N.; Feng, H.; Xu, S.; Wang, B. Org. Lett. 2012, 14, 736. (h) Tan, X.; Liu, B.; Li, X.; Li, B.; Xu, S.; Song, H.; Wang, B. J. Am. Chem. Soc. 2012, 134, 16163. (i) Li, B.; Feng, H.; Xu, S.; Wang, B. Chem. Eur. J. 2011, 17, 12573. (17) For details of the optimization and the NOE experiment of the structure, see the Supporting Information.

(18) (a) Zhang, J.; Loh, T.-P. Chem. Commun. 2012, 48, 11232. (b) Feng, R.; Yu, [W.; Wang, K.; Liu, Z.;](#page-8-0) Zhang, Y. Adv. Synth. Catal. 2014, 356, 1501.

(19) White, C.; Yates, A.; Maitlis, P. M. Inorg. Synth. 1992, 29, 228. (20) Mahipal, B.; Yaragorla, S. R.; Reddy, C. R. Tetrahedron Lett. 2007, 48, 7528.

(21) Matsumura, K.; Hayashi, K.; Saki, K.; Yamamori, T.; Ishizuka, N. Synthesis 2000, 6, 784.

(22) (a) Pham, M. V.; Ye, B.; Cramer, N. Angew. Chem., Int. Ed. 2012, 51, 10610. (b) Fu, S.; Lian, X.; Ma, T.; Chen, W.; Zheng, M.; Zeng, W. Tetrahedron Lett. 2010, 51, 5834. (c) Paulvannan, K.; Stille, J. R. J. Org. Chem. 1992, 57, 5319.